

Heptane, 1,1'-oxybis-

Other names:	1,1'-oxybisheptane
	1-(Heptyloxy)heptane
	Bis(1-heptyl) ether
	Di-n-heptyl ether
	Diheptyl ether
	Ether, di-n-heptyl
	Ether, diheptyl
	Heptyl ether
	NSC 97274
Inchi:	InChI=1S/C14H30O/c1-3-5-7-9-11-13-15-14-12-10-8-6-4-2/h3-14H2,1-2H3
InchiKey:	UJEGHEMJVNQWOJ-UHFFFAOYSA-N
Formula:	C14H30O
SMILES:	CCCCCCCCOCCCCCCC
Mol. weight [g/mol]:	214.39
CAS:	629-64-1

Physical Properties

Property code	Value	Unit	Source
gf	-38.00	kJ/mol	Joback Method
hf	-464.51	kJ/mol	Joback Method
hfus	33.20	kJ/mol	Joback Method
hvap	49.17	kJ/mol	Joback Method
log10ws	-4.77		Crippen Method
logp	4.944		Crippen Method
mcvol	213.990	ml/mol	McGowan Method
pc	1503.48	kPa	Joback Method
rinpol	1458.00		NIST Webbook
ripol	1550.00		NIST Webbook
tb	542.14	K	Joback Method
tc	701.88	K	Joback Method
tf	269.77	K	Joback Method
vc	0.838	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	633.50	J/mol×K	701.88	Joback Method
cpg	536.33	J/mol×K	542.14	Joback Method
cpg	554.10	J/mol×K	568.76	Joback Method
cpg	571.22	J/mol×K	595.39	Joback Method
cpg	587.72	J/mol×K	622.01	Joback Method
cpg	603.58	J/mol×K	648.63	Joback Method
cpg	618.84	J/mol×K	675.26	Joback Method
dvisc	0.0001482	Pa×s	542.14	Joback Method
dvisc	0.0038629	Pa×s	269.77	Joback Method
dvisc	0.0015169	Pa×s	315.16	Joback Method
dvisc	0.0007537	Pa×s	360.56	Joback Method
dvisc	0.0004379	Pa×s	405.95	Joback Method
dvisc	0.0002838	Pa×s	451.35	Joback Method
dvisc	0.0001991	Pa×s	496.75	Joback Method
hvapt	63.10	kJ/mol	453.50	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.61961e+01
Coeff. B	-5.06843e+03
Coeff. C	-8.99220e+01
Temperature range (K), min.	408.52
Temperature range (K), max.	555.57

Sources

The Yaws Handbook of Vapor
Pressure:
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>
<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

https://www.chemeo.com/doc/models/crippen_log10ws

Joback Method:

https://en.wikipedia.org/wiki/Joback_method

KDB: <https://www.cheric.org/files/research/kdb/mol/mol1026.mol>
McGowan Method: <http://link.springer.com/article/10.1007/BF02311772>
NIST Webbook: <http://webbook.nist.gov/cgi/cbook.cgi?ID=C629641&Units=SI>

Legend

cp_g:	Ideal gas heat capacity
dv_{isc}:	Dynamic viscosity
g_f:	Standard Gibbs free energy of formation
h_f:	Enthalpy of formation at standard conditions
h_{fus}:	Enthalpy of fusion at standard conditions
h_{vap}:	Enthalpy of vaporization at standard conditions
h_{vapt}:	Enthalpy of vaporization at a given temperature
log_{10ws}:	Log10 of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
mc_{vol}:	McGowan's characteristic volume
p_c:	Critical Pressure
p_{vap}:	Vapor pressure
ri_{npol}:	Non-polar retention indices
ri_{pol}:	Polar retention indices
t_b:	Normal Boiling Point Temperature
t_c:	Critical Temperature
t_f:	Normal melting (fusion) point
v_c:	Critical Volume

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