

Benzene, 1-iodo-2-methyl-

Other names:	1-Iodo-2-methylbenzene 2-Iodotoluene Toluene, o-iodo- o-Iodotoluene o-Methyliodobenzene o-Tolyl iodide
Inchi:	InChI=1S/C7H7I/c1-6-4-2-3-5-7(6)8/h2-5H,1H3
InchiKey:	RINOYHWVBUKAQE-UHFFFAOYSA-N
Formula:	C7H7I
SMILES:	Cc1cccc1I
Mol. weight [g/mol]:	218.03
CAS:	615-37-2

Physical Properties

Property code	Value	Unit	Source
affp	780.30	kJ/mol	NIST Webbook
basg	750.80	kJ/mol	NIST Webbook
chl	-3833.40 ± 4.20	kJ/mol	NIST Webbook
gf	168.96	kJ/mol	Joback Method
hf	133.00 ± 5.90	kJ/mol	NIST Webbook
hfl	78.20 ± 4.20	kJ/mol	NIST Webbook
hfus	11.94	kJ/mol	Joback Method
hvap	54.40 ± 4.20	kJ/mol	NIST Webbook
ie	8.50 ± 0.10	eV	NIST Webbook
ie	8.62 ± 0.01	eV	NIST Webbook
ie	8.65	eV	NIST Webbook
ie	8.40	eV	NIST Webbook
log10ws	-3.10		Crippen Method
logp	2.600		Crippen Method
mcvol	111.550	ml/mol	McGowan Method
pc	3925.85	kPa	Joback Method
rinpol	1122.40		NIST Webbook
rinpol	1149.00		NIST Webbook
rinpol	1150.00		NIST Webbook
rinpol	1150.00		NIST Webbook
rinpol	1122.40		NIST Webbook
ripol	1749.00		NIST Webbook

ripol	1749.00			NIST Webbook
tb	484.70		K	NIST Webbook
tc	737.63		K	Joback Method
tf	265.65		K	Joback Method
vc	0.407		m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	202.97	J/molxK	568.78	Joback Method
cpg	182.77	J/molxK	484.36	Joback Method
cpg	211.92	J/molxK	610.99	Joback Method
cpg	220.18	J/molxK	653.20	Joback Method
cpg	227.78	J/molxK	695.42	Joback Method
cpg	234.79	J/molxK	737.63	Joback Method
cpg	193.27	J/molxK	526.57	Joback Method
dvisc	0.0016327	Paxs	302.10	Joback Method
dvisc	0.0010247	Paxs	338.55	Joback Method
dvisc	0.0007040	Paxs	375.00	Joback Method
dvisc	0.0005170	Paxs	411.46	Joback Method
dvisc	0.0003992	Paxs	447.91	Joback Method
dvisc	0.0003205	Paxs	484.36	Joback Method
dvisc	0.0029563	Paxs	265.65	Joback Method
hvapt	49.70	kJ/mol	397.00	NIST Webbook
pvap	0.28	kPa	325.10	Thermochemistry of Halogen-Substituted Methylbenzenes
pvap	0.23	kPa	322.10	Thermochemistry of Halogen-Substituted Methylbenzenes
pvap	0.20	kPa	319.30	Thermochemistry of Halogen-Substituted Methylbenzenes
pvap	0.17	kPa	316.20	Thermochemistry of Halogen-Substituted Methylbenzenes
pvap	0.14	kPa	313.10	Thermochemistry of Halogen-Substituted Methylbenzenes

pvap	0.11	kPa	310.30	Thermochemistry of Halogen-Substituted Methylbenzenes
pvap	0.09	kPa	307.40	Thermochemistry of Halogen-Substituted Methylbenzenes
pvap	0.08	kPa	304.20	Thermochemistry of Halogen-Substituted Methylbenzenes
pvap	0.06	kPa	301.20	Thermochemistry of Halogen-Substituted Methylbenzenes
pvap	0.04	kPa	293.20	Thermochemistry of Halogen-Substituted Methylbenzenes
pvap	0.02	kPa	288.20	Thermochemistry of Halogen-Substituted Methylbenzenes
pvap	0.02	kPa	283.20	Thermochemistry of Halogen-Substituted Methylbenzenes

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.39088e+01
Coeff. B	-3.79809e+03
Coeff. C	-7.58830e+01
Temperature range (K), min.	354.72
Temperature range (K), max.	517.66

Sources

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C615372&Units=SI>

The Yaws Handbook of Vapor Pressure:
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Thermochemistry of Halogen-Substituted Methylbenzenes:	https://www.doi.org/10.1021/je500784s
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

affp:	Proton affinity
basg:	Gas basicity
chl:	Standard liquid enthalpy of combustion
cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfl:	Liquid phase enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
rinpol:	Non-polar retention indices
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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<https://www.chemeo.com/cid/21-410-2/Benzene-1-iodo-2-methyl.pdf>

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