

# Borane carbonyl

<b>Other names:</b>	BH3CO Borane, compd. with carbon monoxide (1:1) Borane-carbon monoxide (1:1) Borine carbonyl (BH3CO) Boron, carbonyltrihydro- Boron, carbonyltrihydro-, (T-4)- Carbon monoxide-borane
<b>Inchi:</b>	InChI=1S/CH3BO/c2-1-3/h2H3
<b>InchiKey:</b>	ZJUVLBOEAXFSHP-UHFFFAOYSA-N
<b>Formula:</b>	CH3BO
<b>SMILES:</b>	[BH3-]C#[O+]
<b>Mol. weight [g/mol]:</b>	41.84
<b>CAS:</b>	13205-44-2

## Physical Properties

Property code	Value	Unit	Source
ie	11.14 ± 0.02	eV	NIST Webbook
ie	11.92 ± 0.02	eV	NIST Webbook
log10ws	-1.42		Crippen Method
logp	-1.301		Crippen Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
hvapt	19.70	kJ/mol	171.50	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$

Coeff. A	1.64002e+01
Coeff. B	-2.51030e+03
Coeff. C	3.91400e+00
Temperature range (K), min.	149.15
Temperature range (K), max.	222.47

## Sources

The Yaws Handbook of Vapor Pressure:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C13205442&Units=SI>

## Legend

**hvapt:** Enthalpy of vaporization at a given temperature

**ie:** Ionization energy

**log10ws:** Log10 of Water solubility in mol/l

**logp:** Octanol/Water partition coefficient

**pvap:** Vapor pressure

Latest version available from:

<https://www.chemeo.com/cid/21-417-5/Borane-carbonyl.pdf>

Generated by Cheméo on 2024-04-25 07:00:13.751449556 +0000 UTC m=+16317662.672026868.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.