

2,2'-Bioxirane

Other names:	Butane, 1,2:3,4-diepoxy-
	Bioxirane
	Butadiene diepoxide
	Butadiene dioxide
	Butane diepoxide
	Diepoxybutane
	Dioxybutadiene
	1,1'-Bi[ethylene oxide]
	1,2,3,4-Diepoxybutane
	1,2:3,4-Diepoxybutane
	Bioxiran
	CB 1181
	M 8838
	R 181
	Threitol, 1,2:3,4-dianhydro-
	Butadiendioxyd
	1,3-Butadiene diepoxide
	DEB
	ENT-26592
	Rcra waste number U085
	NSC 629
Inchi:	InChI=1S/C4H6O2/c1-3(5-1)4-2-6-4/h3-4H,1-2H2
InchiKey:	ZFIVKAOQEXOYFY-UHFFFAOYSA-N
Formula:	C4H6O2
SMILES:	C1OC1C1CO1
Mol. weight [g/mol]:	86.09
CAS:	1464-53-5

Physical Properties

Property code	Value	Unit	Source
gf	-67.94	kJ/mol	Joback Method
hf	-244.29	kJ/mol	Joback Method
hfus	18.34	kJ/mol	Joback Method
hvap	33.34	kJ/mol	Joback Method
log10ws	0.32		Crippen Method
logp	-0.216		Crippen Method

mcvol	57.240	ml/mol	McGowan Method
pc	5406.57	kPa	Joback Method
tb	358.30	K	Joback Method
tc	558.39	K	Joback Method
tf	223.86	K	Joback Method
vc	0.215	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	109.07	J/mol×K	358.30	Joback Method
cpg	119.63	J/mol×K	391.65	Joback Method
cpg	129.41	J/mol×K	425.00	Joback Method
cpg	138.46	J/mol×K	458.35	Joback Method
cpg	146.83	J/mol×K	491.69	Joback Method
cpg	154.56	J/mol×K	525.04	Joback Method
cpg	161.70	J/mol×K	558.39	Joback Method
dvisc	0.0007038	Pa×s	223.86	Joback Method
dvisc	0.0006732	Pa×s	246.27	Joback Method
dvisc	0.0006488	Pa×s	268.67	Joback Method
dvisc	0.0006288	Pa×s	291.08	Joback Method
dvisc	0.0006122	Pa×s	313.49	Joback Method
dvisc	0.0005981	Pa×s	335.89	Joback Method
dvisc	0.0005861	Pa×s	358.30	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C1464535&Units=SI

Legend

cpg: Ideal gas heat capacity

dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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