

# Phenol, 2-(1,1-dimethylethyl)-4-methyl-

<b>Other names:</b>	1-Hydroxy-2-tert-butyl-4-methylbenzene 2-(1,1'-dimethylethyl)-4-methylphenol 2-(1,1-Dimethylethyl)-4-methyl-phenol 2-Terc.butyl-p-kresol 2-t-Butyl-4-methylphenol 2-t-Butyl-p-cresol 2-tert-Butyl-4-Methylphenol 2-tert-Butyl-4-cresol 2-tert-Butyl-4-methyl-1-phenol 2-tert-Butyl-p-Cresol 4-Methyl-2-(1,1-dimethylethyl)phenol 4-Methyl-2-tert-butylphenol 4-Methyl-6-t-butylphenol NSC 60301 o-tert-Butyl-p-cresol p-Cresol, 2-tert-butyl-
<b>Inchi:</b>	InChI=1S/C11H16O/c1-8-5-6-10(12)9(7-8)11(2,3)4/h5-7,12H,1-4H3
<b>InchiKey:</b>	IKEHOXWJQXIQAG-UHFFFAOYSA-N
<b>Formula:</b>	C11H16O
<b>SMILES:</b>	<chem>Cc1ccc(O)c(C(C)(C)C)c1</chem>
<b>Mol. weight [g/mol]:</b>	164.24
<b>CAS:</b>	2409-55-4

## Physical Properties

Property code	Value	Unit	Source
gf	-7.26	kJ/mol	Joback Method
hf	-207.00 ± 3.20	kJ/mol	NIST Webbook
hfus	16.27	kJ/mol	Joback Method
hsub	82.90 ± 0.50	kJ/mol	NIST Webbook
hvap	65.70 ± 0.30	kJ/mol	NIST Webbook
log10ws	-2.81		Crippen Method
logp	2.998		Crippen Method
mcvol	147.960	ml/mol	McGowan Method
pc	3121.00	kPa	Joback Method
rinpol	1352.20		NIST Webbook
rinpol	1355.00		NIST Webbook
rinpol	1353.10		NIST Webbook

rinpol	1387.00		NIST Webbook
rinpol	1352.20		NIST Webbook
rinpol	1345.00		NIST Webbook
ripol	2235.00		NIST Webbook
ripol	2235.00		NIST Webbook
tb	510.20	K	NIST Webbook
tc	789.71	K	Joback Method
tf	325.16 ± 0.30	K	NIST Webbook
tf	325.45 ± 0.20	K	NIST Webbook
vc	0.498	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	360.74	J/mol×K	560.13	Joback Method
cpg	375.98	J/mol×K	598.39	Joback Method
cpg	390.09	J/mol×K	636.66	Joback Method
cpg	403.20	J/mol×K	674.92	Joback Method
cpg	415.40	J/mol×K	713.18	Joback Method
cpg	426.81	J/mol×K	751.44	Joback Method
cpg	437.54	J/mol×K	789.71	Joback Method
dvisc	0.0000425	Paxs	560.13	Joback Method
dvisc	0.0008050	Paxs	399.03	Joback Method
dvisc	0.0003749	Paxs	431.25	Joback Method
dvisc	0.0001942	Paxs	463.47	Joback Method
dvisc	0.0019771	Paxs	366.81	Joback Method
dvisc	0.0000663	Paxs	527.91	Joback Method
dvisc	0.0001095	Paxs	495.69	Joback Method
hsubt	82.60 ± 0.50	kJ/mol	303.00	NIST Webbook
hsubt	77.40	kJ/mol	284.00	NIST Webbook
hvapt	63.00 ± 0.30	kJ/mol	342.50	NIST Webbook
hvapt	58.90	kJ/mol	451.00	NIST Webbook
hvapt	57.70	kJ/mol	425.00	NIST Webbook
hvapt	55.70	kJ/mol	425.00	NIST Webbook
hvapt	52.60	kJ/mol	425.00	NIST Webbook
hvapt	48.50	kJ/mol	425.00	NIST Webbook

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.44886e+01
Coeff. B	-4.33342e+03
Coeff. C	-8.29720e+01
Temperature range (K), min.	388.12
Temperature range (K), max.	555.17

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2409554&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2409554&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure

<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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