

# [1,1'-Biphenyl]-4-carbonitrile, 4'-(octyloxy)-

<b>Other names:</b>	4-Cyano-4'-(octyloxy)-1,1'-biphenyl p-Cyano-p'-(octyloxy)biphenyl 4'-(n-Octyloxy)-4-cyanobiphenyl 4-Cyano-4'-octoxybiphenyl 4-(Octyloxy)-4'-cyanobiphenyl 4-Cyano-4'-(n-octyloxy)biphenyl 4'-(Octyloxy)-4-cyanobiphenyl p-(Octyloxy)-p'-cyanobiphenyl 4'-(Octyloxy)-4-biphenylcarbonitrile 4-(4-Octyloxyphenyl)benzonitrile 8OCB M 24 M 24 (liquid crystal) 4'-(octyloxy)[1,1'-biphenyl]-4-carbonitrile Octyloxycyanobiphenyl
<b>Inchi:</b>	InChI=1S/C21H25NO/c1-2-3-4-5-6-7-16-23-21-14-12-20(13-15-21)19-10-8-18(17-22)9-1
<b>InchiKey:</b>	GPGGNNIMKOVSAU-UHFFFAOYSA-N
<b>Formula:</b>	C21H25NO
<b>SMILES:</b>	CCCCCCCCOc1ccc(-c2ccc(C#N)cc2)cc1
<b>Mol. weight [g/mol]:</b>	307.43
<b>CAS:</b>	52364-73-5

## Physical Properties

Property code	Value	Unit	Source
gf	359.68	kJ/mol	Joback Method
hf	6.01	kJ/mol	Joback Method
hfus	40.14	kJ/mol	Joback Method
hvap	81.10	kJ/mol	Joback Method
log10ws	-7.48		Crippen Method
logp	5.965		Crippen Method
mvol	266.480	ml/mol	McGowan Method
pc	1430.46	kPa	Joback Method
tb	867.70	K	Joback Method
tc	1091.14	K	Joback Method
tf	327.70 ± 0.20	K	NIST Webbook
vc	1.040	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	805.36	J/mol×K	867.70	Joback Method
cpg	820.67	J/mol×K	904.94	Joback Method
cpg	834.81	J/mol×K	942.18	Joback Method
cpg	847.84	J/mol×K	979.42	Joback Method
cpg	859.82	J/mol×K	1016.66	Joback Method
cpg	870.80	J/mol×K	1053.90	Joback Method
cpg	880.84	J/mol×K	1091.14	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C52364735&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C52364735&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/21-486-9/1-1-Biphenyl-4-carbonitrile-4-octyloxy.pdf>

Generated by Cheméo on 2024-04-27 14:54:18.369379794 +0000 UTC m=+16518907.289957109.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.