

# 4-Propylcyclohexanone

<b>Other names:</b>	Cyclohexanone, 4-propyl-
<b>Inchi:</b>	InChI=1S/C9H16O/c1-2-3-8-4-6-9(10)7-5-8/h8H,2-7H2,1H3
<b>InchiKey:</b>	NQEDLIZOPMNZMC-UHFFFAOYSA-N
<b>Formula:</b>	C9H16O
<b>SMILES:</b>	CCCC1CCC(=O)CC1
<b>Mol. weight [g/mol]:</b>	140.22
<b>CAS:</b>	40649-36-3

## Physical Properties

Property code	Value	Unit	Source
gf	-73.24	kJ/mol	Joback Method
hf	-312.47	kJ/mol	Joback Method
hfus	10.41	kJ/mol	Joback Method
hvap	40.30	kJ/mol	Joback Method
log10ws	-2.52		Crippen Method
logp	2.546		Crippen Method
mcvol	128.380	ml/mol	McGowan Method
pc	2947.28	kPa	Joback Method
tb	492.69	K	Joback Method
tc	709.61	K	Joback Method
tf	266.79	K	Joback Method
vc	0.479	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	291.50	J/mol×K	492.69	Joback Method
cpg	309.49	J/mol×K	528.84	Joback Method
cpg	326.65	J/mol×K	565.00	Joback Method
cpg	342.98	J/mol×K	601.15	Joback Method
cpg	358.49	J/mol×K	637.31	Joback Method
cpg	373.17	J/mol×K	673.46	Joback Method
cpg	387.02	J/mol×K	709.61	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	370.70	K	2.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.60147e+01
Coeff. B	-4.48569e+03
Coeff. C	-7.63000e+01
Temperature range (K), min.	361.52
Temperature range (K), max.	495.40

## Sources

The Yaws Handbook of Vapor Pressure:	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C40649363&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C40649363&amp;Units=SI</a>

## Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions

<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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