

# 1-Propene, 2-methoxy-

<b>Other names:</b>	Ether, isopropenyl methyl Isopropenyl methyl ether Methyl isopropenyl ether 2-Methoxy-1-propene <chem>CH2=C(CH3)OCH3</chem> 2-Methoxypropene
<b>Inchi:</b>	InChI=1S/C4H8O/c1-4(2)5-3/h1H2,2-3H3
<b>InchiKey:</b>	YOWQWFMSQCOSBA-UHFFFAOYSA-N
<b>Formula:</b>	C4H8O
<b>SMILES:</b>	<chem>C=C(C)OC</chem>
<b>Mol. weight [g/mol]:</b>	72.11
<b>CAS:</b>	116-11-0

## Physical Properties

Property code	Value	Unit	Source
affp	894.90	kJ/mol	NIST Webbook
basg	866.10	kJ/mol	NIST Webbook
gf	-42.91	kJ/mol	Joback Method
hf	-148.60	kJ/mol	NIST Webbook
hfus	4.71	kJ/mol	Joback Method
hvap	26.32	kJ/mol	Joback Method
ie	8.64	eV	NIST Webbook
log10ws	-0.93		Crippen Method
logp	1.166		Crippen Method
mcvol	68.790	ml/mol	McGowan Method
pc	4046.64	kPa	Joback Method
rinpol	482.00		NIST Webbook
rinpol	482.00		NIST Webbook
rinpol	483.00		NIST Webbook
tb	308.20	K	NIST Webbook
tc	480.47	K	Joback Method
tf	141.35	K	Joback Method
vc	0.260	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	104.86	J/mol×K	309.90	Joback Method
cpg	111.42	J/mol×K	338.33	Joback Method
cpg	117.81	J/mol×K	366.76	Joback Method
cpg	124.02	J/mol×K	395.18	Joback Method
cpg	130.06	J/mol×K	423.61	Joback Method
cpg	135.93	J/mol×K	452.04	Joback Method
cpg	141.63	J/mol×K	480.47	Joback Method
cpl	162.20	J/mol×K	301.50	NIST Webbook
hvapt	28.30 ± 0.10	kJ/mol	295.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C116110&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C116110&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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