

1-Propene, 2-methoxy-

Other names:	Ether, isopropenyl methyl Isopropenyl methyl ether Methyl isopropenyl ether 2-Methoxy-1-propene <chem>CH2=C(CH3)OCH3</chem> 2-Methoxypropene
Inchi:	InChI=1S/C4H8O/c1-4(2)5-3/h1H2,2-3H3
InchiKey:	YOWQWFMSQCOSBA-UHFFFAOYSA-N
Formula:	C4H8O
SMILES:	<chem>C=C(C)OC</chem>
Mol. weight [g/mol]:	72.11
CAS:	116-11-0

Physical Properties

Property code	Value	Unit	Source
affp	894.90	kJ/mol	NIST Webbook
basg	866.10	kJ/mol	NIST Webbook
gf	-42.91	kJ/mol	Joback Method
hf	-148.60	kJ/mol	NIST Webbook
hfus	4.71	kJ/mol	Joback Method
hvap	26.32	kJ/mol	Joback Method
ie	8.64	eV	NIST Webbook
log10ws	-0.93		Crippen Method
logp	1.166		Crippen Method
mcvol	68.790	ml/mol	McGowan Method
pc	4046.64	kPa	Joback Method
rinpol	482.00		NIST Webbook
rinpol	483.00		NIST Webbook
rinpol	482.00		NIST Webbook
tb	308.20	K	NIST Webbook
tc	480.47	K	Joback Method
tf	141.35	K	Joback Method
vc	0.260	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	104.86	J/mol×K	309.90	Joback Method
cpg	111.42	J/mol×K	338.33	Joback Method
cpg	117.81	J/mol×K	366.76	Joback Method
cpg	124.02	J/mol×K	395.18	Joback Method
cpg	130.06	J/mol×K	423.61	Joback Method
cpg	135.93	J/mol×K	452.04	Joback Method
cpg	141.63	J/mol×K	480.47	Joback Method
cpl	162.20	J/mol×K	301.50	NIST Webbook
hvapt	28.30 ± 0.10	kJ/mol	295.00	NIST Webbook

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C116110&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

affp:	Proton affinity
basg:	Gas basicity
cpg:	Ideal gas heat capacity
cpl:	Liquid phase heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mconvol:	McGowan's characteristic volume

pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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