

# Phenol, 2,3-dimethyl-

<b>Other names:</b>	1,2-Dimethyl-3-hydroxybenzene 1-HYDROXY-2,3-DIMETHYLBENZENE 2,3-DIMETHYLPHENOL 2,3-Xylenol NSC 62011 o-Xylenol
<b>Inchi:</b>	InChI=1S/C8H10O/c1-6-4-3-5-8(9)7(6)2/h3-5,9H,1-2H3
<b>InchiKey:</b>	QWBBPBRQALCEIZ-UHFFFAOYSA-N
<b>Formula:</b>	C8H10O
<b>SMILES:</b>	Cc1cccc(O)c1C
<b>Mol. weight [g/mol]:</b>	122.16
<b>CAS:</b>	526-75-0

## Physical Properties

Property code	Value	Unit	Source
chs	-4336.00 ± 0.92	kJ/mol	NIST Webbook
gf	-35.36	kJ/mol	Joback Method
hf	-157.30 ± 1.10	kJ/mol	NIST Webbook
hf	-157.30	kJ/mol	KDB
hfs	-241.30 ± 1.10	kJ/mol	NIST Webbook
hfus	15.91	kJ/mol	Joback Method
hsub	84.00 ± 1.00	kJ/mol	NIST Webbook
hsub	84.00	kJ/mol	NIST Webbook
hsub	59.40 ± 0.80	kJ/mol	NIST Webbook
hvap	49.35	kJ/mol	Joback Method
ie	8.26	eV	NIST Webbook
log10ws	-1.43		Aqueous Solubility Prediction Method
logp	2.009		Crippen Method
mvol	105.690	ml/mol	McGowan Method
pc	4322.57	kPa	Joback Method
rinpol	1147.00		NIST Webbook
rinpol	1153.00		NIST Webbook
rinpol	198.95		NIST Webbook
rinpol	198.65		NIST Webbook
rinpol	198.95		NIST Webbook
rinpol	1167.00		NIST Webbook

rinpol	1169.00		NIST Webbook
rinpol	1181.00		NIST Webbook
rinpol	1178.00		NIST Webbook
rinpol	1169.00		NIST Webbook
rinpol	1132.00		NIST Webbook
rinpol	196.80		NIST Webbook
rinpol	1132.00		NIST Webbook
rinpol	1150.00		NIST Webbook
rinpol	1159.00		NIST Webbook
rinpol	1159.00		NIST Webbook
rinpol	1159.00		NIST Webbook
rinpol	1160.00		NIST Webbook
rinpol	1143.00		NIST Webbook
rinpol	1143.00		NIST Webbook
rinpol	1159.00		NIST Webbook
rinpol	1132.00		NIST Webbook
rinpol	1169.00		NIST Webbook
rinpol	1151.00		NIST Webbook
rinpol	1181.00		NIST Webbook
rinpol	1190.60		NIST Webbook
rinpol	1178.90		NIST Webbook
rinpol	1180.10		NIST Webbook
rinpol	1182.20		NIST Webbook
rinpol	1184.00		NIST Webbook
rinpol	1153.00		NIST Webbook
rinpol	1200.00		NIST Webbook
rinpol	1132.00		NIST Webbook
rinpol	1151.00		NIST Webbook
rinpol	1166.00		NIST Webbook
rinpol	1167.00		NIST Webbook
rinpol	1158.00		NIST Webbook
rinpol	198.03		NIST Webbook
ripol	2147.00		NIST Webbook
ripol	2155.00		NIST Webbook
ripol	2143.00		NIST Webbook
ripol	2137.00		NIST Webbook
ripol	2140.00		NIST Webbook
ripol	2147.00		NIST Webbook
ripol	2120.00		NIST Webbook
ripol	2188.00		NIST Webbook
ripol	2147.00		NIST Webbook
tb	491.15 ± 2.00	K	NIST Webbook
tb	490.07	K	KDB
tb	491.20 ± 2.00	K	NIST Webbook

tb	491.15 ± 2.00	K	NIST Webbook
tb	490.95 ± 2.00	K	NIST Webbook
tb	491.15 ± 3.00	K	NIST Webbook
tb	490.25 ± 2.00	K	NIST Webbook
tb	492.15 ± 3.00	K	NIST Webbook
tb	491.20	K	NIST Webbook
tb	484.65 ± 5.00	K	NIST Webbook
tc	722.85 ± 0.40	K	NIST Webbook
tc	722.80	K	KDB
tf	345.91	K	Aqueous Solubility Prediction Method
tf	345.90	K	KDB
vc	0.342	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	282.81	J/mol×K	722.03	Joback Method
cpg	224.69	J/mol×K	494.72	Joback Method
cpg	236.18	J/mol×K	532.60	Joback Method
cpg	246.85	J/mol×K	570.49	Joback Method
cpg	256.78	J/mol×K	608.37	Joback Method
cpg	266.04	J/mol×K	646.26	Joback Method
cpg	274.69	J/mol×K	684.14	Joback Method
dvisc	0.0000891	Paxs	494.72	Joback Method
dvisc	0.0031710	Paxs	330.58	Joback Method
dvisc	0.0013928	Paxs	357.94	Joback Method
dvisc	0.0006876	Paxs	385.29	Joback Method
dvisc	0.0003727	Paxs	412.65	Joback Method
dvisc	0.0002180	Paxs	440.01	Joback Method
dvisc	0.0001358	Paxs	467.36	Joback Method
hfust	21.02	kJ/mol	346.00	NIST Webbook
hfust	21.02	kJ/mol	346.00	NIST Webbook
hfust	20.29	kJ/mol	345.80	NIST Webbook
hfust	21.02	kJ/mol	346.00	NIST Webbook
hsubt	61.50 ± 0.80	kJ/mol	310.00	NIST Webbook
hsubt	84.00 ± 1.00	kJ/mol	303.00	NIST Webbook
hvapt	52.10	kJ/mol	462.50	NIST Webbook
sfust	60.76	J/mol×K	346.00	NIST Webbook

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.54381e+01
Coeff. B	-4.46433e+03
Coeff. C	-7.73890e+01
Temperature range (K), min.	345.71
Temperature range (K), max.	518.24

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	1.05582e+02
Coeff. B	-1.07419e+04
Coeff. C	-1.30204e+01
Coeff. D	6.74763e-06
Temperature range (K), min.	345.71
Temperature range (K), max.	722.95

# Sources

<b>Solubilities of Substituted Phenols in Supercritical Carbon Dioxide:</b>	<a href="https://www.doi.org/10.1021/je060058e">https://www.doi.org/10.1021/je060058e</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.chemic.org/research/kdb/hcprop/showprop.php?cmpid=874">https://www.chemic.org/research/kdb/hcprop/showprop.php?cmpid=874</a>
<b>Experiments and COSMO-SAC Modeling of Methyl Isobutyl Ketone + The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.doi.org/10.1021/acs.jced.9b00300">https://www.doi.org/10.1021/acs.jced.9b00300</a>
<b>Formation of Deep Eutectic Solvents by Phenols and Choline Chloride and Their Physical Properties:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Joback Method:</b>	<a href="https://www.doi.org/10.1021/je300997v">https://www.doi.org/10.1021/je300997v</a>
<b>McGowan Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="https://www.chemic.org/files/research/kdb/mol/mol874.mol">https://www.chemic.org/files/research/kdb/mol/mol874.mol</a>
	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C526750&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C526750&amp;Units=SI</a>

# Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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