

# Methyl n-butyl disulfide

<b>Other names:</b>	1-(Methyldisulfanyl)butane 2,3-Diathiaheptane 2,3-Dithiaheptane Butyl methyl disulfide Disulfide, butyl methyl Methyl butyl disulfide Methyl n-butyl disulphide
<b>Inchi:</b>	InChI=1S/C5H12S2/c1-3-4-5-7-6-2/h3-5H2,1-2H3
<b>InchiKey:</b>	ABUPWXCYFWRZKL-UHFFFAOYSA-N
<b>Formula:</b>	C5H12S2
<b>SMILES:</b>	CCCCSSC
<b>Mol. weight [g/mol]:</b>	136.28
<b>CAS:</b>	60779-24-0

## Physical Properties

Property code	Value	Unit	Source
gf	57.46	kJ/mol	Joback Method
hf	-62.79	kJ/mol	Joback Method
hfus	16.97	kJ/mol	Joback Method
hvap	40.36	kJ/mol	Joback Method
log10ws	-2.67		Crippen Method
logp	2.798		Crippen Method
mcvol	114.010	ml/mol	McGowan Method
pc	3611.55	kPa	Joback Method
rinpol	1016.00		NIST Webbook
rinpol	1044.00		NIST Webbook
rinpol	1044.00		NIST Webbook
rinpol	1044.00		NIST Webbook
rinpol	986.00		NIST Webbook
ripol	1351.00		NIST Webbook
ripol	1366.00		NIST Webbook
ripol	1351.00		NIST Webbook
ripol	1366.00		NIST Webbook
tb	451.36	K	Joback Method
tc	666.63	K	Joback Method
tf	214.91	K	Joback Method
vc	0.423	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	213.79	J/mol×K	451.36	Joback Method
cpg	224.75	J/mol×K	487.24	Joback Method
cpg	235.26	J/mol×K	523.12	Joback Method
cpg	245.30	J/mol×K	558.99	Joback Method
cpg	254.88	J/mol×K	594.87	Joback Method
cpg	264.00	J/mol×K	630.75	Joback Method
cpg	272.66	J/mol×K	666.63	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.52527e+01
Coeff. B	-4.01255e+03
Coeff. C	-6.64510e+01
Temperature range (K), min.	334.58
Temperature range (K), max.	470.08

## Sources

The Yaws Handbook of Vapor Pressure:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>  
<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C60779240&Units=SI>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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