

# («alpha»,«alpha»,«alpha»-Trifluoro-p-tolyl)-acetic acid

Other names:  
**acid**

4-Trifluoromethylphenylacetic acid

p-Trifluoromethylphenylacetic acid

Benzeneacetic acid, 4-(trifluoromethyl)-

(«alpha»,«alpha»,«alpha»-trifluoro-4-tolyl)acetic acid

**Inchi:** InChI=1S/C9H7F3O2/c10-9(11,12)7-3-1-6(2-4-7)5-8(13)14/h1-4H,5H2,(H,13,14)

**InchiKey:** HNORVZDAANCHAY-UHFFFAOYSA-N

**Formula:** C9H7F3O2

**SMILES:** O=C(O)Cc1ccc(C(F)(F)F)cc1

**Mol. weight [g/mol]:** 204.15

**CAS:** 32857-62-8

## Physical Properties

Property code	Value	Unit	Source
gf	-719.65	kJ/mol	Joback Method
hf	-865.92	kJ/mol	Joback Method
hfus	20.23	kJ/mol	Joback Method
hvap	58.24	kJ/mol	Joback Method
log10ws	-2.48		Crippen Method
logp	2.333		Crippen Method
mcvol	126.660	ml/mol	McGowan Method
pc	3352.86	kPa	Joback Method
tb	577.61	K	Joback Method
tc	765.80	K	Joback Method
tf	345.07	K	Joback Method
vc	0.499	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	308.99	J/molxK	577.61	Joback Method
cpg	318.33	J/molxK	608.98	Joback Method
cpg	327.03	J/molxK	640.34	Joback Method
cpg	335.13	J/molxK	671.71	Joback Method
cpg	342.67	J/molxK	703.07	Joback Method

cpg	349.67	J/mol×K	734.44	Joback Method
cpg	356.17	J/mol×K	765.80	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C32857628&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C32857628&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcpv:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/21-645-2/alpha-alpha-alpha-Trifluoro-p-tolyl-acetic-acid.pdf>

Generated by Cheméo on 2024-04-23 12:46:10.075565246 +0000 UTC m=+16165618.996142558.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.