

# 5-Hepten-2-amine, N,6-dimethyl-

<b>Other names:</b>	4-Hexenylamine, N,1,5-trimethyl- Isometheptene Isonyl 6-Methylamino-2-methyl-2-heptene 2-Methyl-6-methylamino-2-heptene Octanil Octin Octinum Octon N,1,5-Trimethyl-4-hexenylamine 6-Methylamino-2-methylheptene Octine
<b>Inchi:</b>	InChI=1S/C9H19N/c1-8(2)6-5-7-9(3)10-4/h6,9-10H,5,7H2,1-4H3
<b>InchiKey:</b>	XVQUOJBERHHONY-UHFFFAOYSA-N
<b>Formula:</b>	C9H19N
<b>SMILES:</b>	CNC(C)CCC=C(C)C
<b>Mol. weight [g/mol]:</b>	141.25
<b>CAS:</b>	503-01-5

## Physical Properties

Property code	Value	Unit	Source
gf	183.52	kJ/mol	Joback Method
hf	-73.47	kJ/mol	Joback Method
hfus	19.53	kJ/mol	Joback Method
hvap	41.71	kJ/mol	Joback Method
log10ws	-2.74		Crippen Method
logp	2.341		Crippen Method
mcvol	143.350	ml/mol	McGowan Method
pc	2490.03	kPa	Joback Method
rinpol	1052.00		NIST Webbook
rinpol	1052.00		NIST Webbook
rinpol	1023.00		NIST Webbook
rinpol	1015.00		NIST Webbook
rinpol	1055.70		NIST Webbook
rinpol	1052.00		NIST Webbook
rinpol	1033.00		NIST Webbook
rinpol	1016.00		NIST Webbook

rinpol	1016.00		NIST Webbook
rinpol	1054.10		NIST Webbook
rinpol	1029.00		NIST Webbook
ripol	1292.00		NIST Webbook
ripol	1289.00		NIST Webbook
ripol	1280.00		NIST Webbook
tb	459.09	K	Joback Method
tc	641.36	K	Joback Method
tf	209.81	K	Joback Method
vc	0.549	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	306.84	J/mol×K	459.09	Joback Method
cpg	321.82	J/mol×K	489.47	Joback Method
cpg	336.10	J/mol×K	519.85	Joback Method
cpg	349.71	J/mol×K	550.22	Joback Method
cpg	362.68	J/mol×K	580.60	Joback Method
cpg	375.04	J/mol×K	610.98	Joback Method
cpg	386.80	J/mol×K	641.36	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C503015&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C503015&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>r<sub>inpol</sub>:</b>	Non-polar retention indices
<b>r<sub>ipol</sub>:</b>	Polar retention indices
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature
<b>t<sub>f</sub>:</b>	Normal melting (fusion) point
<b>v<sub>c</sub>:</b>	Critical Volume

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