

# Cumyl bromide

<b>Inchi:</b>	InChI=1S/C9H11Br/c1-9(2,10)8-6-4-3-5-7-8/h3-7H,1-2H3
<b>InchiKey:</b>	WSFSJXBURLJOFD-UHFFFAOYSA-N
<b>Formula:</b>	C9H11Br
<b>SMILES:</b>	CC(C)(Br)c1ccccc1
<b>Mol. weight [g/mol]:</b>	199.09

## Physical Properties

Property code	Value	Unit	Source
gf	154.47	kJ/mol	Joback Method
hf	25.02	kJ/mol	Joback Method
hfus	10.98	kJ/mol	Joback Method
hvap	43.04	kJ/mol	Joback Method
log10ws	-3.29		Crippen Method
logp	3.317		Crippen Method
mcvol	131.410	ml/mol	McGowan Method
pc	3615.89	kPa	Joback Method
rinsol	1181.00		NIST Webbook
tb	494.93	K	Joback Method
tc	735.23	K	Joback Method
tf	279.83	K	Joback Method
vc	0.482	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	257.96	J/mol×K	494.93	Joback Method
cpg	319.41	J/mol×K	695.18	Joback Method
cpg	309.17	J/mol×K	655.13	Joback Method
cpg	298.00	J/mol×K	615.08	Joback Method
cpg	285.80	J/mol×K	575.03	Joback Method
cpg	272.48	J/mol×K	534.98	Joback Method
cpg	328.79	J/mol×K	735.23	Joback Method
dvisc	0.0002728	Paxs	494.93	Joback Method
dvisc	0.0003576	Paxs	459.08	Joback Method

dvisc	0.0004909	Paxs	423.23	Joback Method
dvisc	0.0007145	Paxs	387.38	Joback Method
dvisc	0.0011228	Paxs	351.53	Joback Method
dvisc	0.0019550	Paxs	315.68	Joback Method
dvisc	0.0039237	Paxs	279.83	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R558897&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R558897&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/21-724-4/Cumyl-bromide.pdf>

Generated by Cheméo on 2024-04-24 01:43:44.30039965 +0000 UTC m=+16212273.220976963.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.