

BENZENE, 1-(CHLOROMETHYL)-2,4-DIMETHYL-

Other names:	1,3-Dimethyl-4-(chloromethyl)benzene 1-(Chloromethyl)-2,4-dimethylbenzene 2,4-Dimethyl-1-(chloromethyl)benzene 4-(Chloromethyl)-m-xylene
Inchi:	InChI=1S/C9H11Cl/c1-7-3-4-9(6-10)8(2)5-7/h3-5H,6H2,1-2H3
InchiKey:	BETNPSBTDMBHCHZ-UHFFFAOYSA-N
Formula:	C9H11Cl
SMILES:	Cc1ccc(CCl)c(C)c1
Mol. weight [g/mol]:	154.64
CAS:	824-55-5

Physical Properties

Property code	Value	Unit	Source
af	0.4010		Cheméo Relay (1.0)
gf	106.12	kJ/mol	Joback Method
hf	-53.22	kJ/mol	Cheméo Relay (1.0)
hfus	16.53	kJ/mol	Joback Method
hvap	56.23	kJ/mol	Cheméo Relay (1.0)
ie	8.61	eV	Cheméo Relay (1.0)
log10ws	-3.51		Cheméo Relay (1.0)
logp	3.042		Crippen Method
mcvol	126.150	ml/mol	McGowan Method
pc	3025.61	kPa	Joback Method
rinpol	1207.00		NIST Webbook
rinpol	1208.00		NIST Webbook
rinpol	1207.00		NIST Webbook
tb	497.26	K	Cheméo Relay (1.0)
tc	710.73	K	Cheméo Relay (1.0)
tf	284.41	K	Cheméo Relay (1.0)
vc	0.459	m3/kmol	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	243.19	J/mol×K	479.39	Joback Method
cpg	255.67	J/mol×K	515.44	Joback Method
cpg	267.48	J/mol×K	551.49	Joback Method
cpg	278.64	J/mol×K	587.55	Joback Method
cpg	289.18	J/mol×K	623.60	Joback Method
cpg	299.10	J/mol×K	659.65	Joback Method
cpg	308.45	J/mol×K	695.70	Joback Method
dvisc	0.0017313	Pa×s	272.57	Joback Method
dvisc	0.0010303	Pa×s	307.04	Joback Method
dvisc	0.0006809	Pa×s	341.51	Joback Method
dvisc	0.0004855	Pa×s	375.98	Joback Method
dvisc	0.0003664	Pa×s	410.45	Joback Method
dvisc	0.0002888	Pa×s	444.92	Joback Method
dvisc	0.0002356	Pa×s	479.39	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.43358e+01
Coeff. B	-3.98863e+03
Coeff. C	-7.69950e+01
Temperature range (K), min.	360.92
Temperature range (K), max.	518.98

Sources

The Yaws Handbook of Vapor Pressure:
NIST Webbook:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>
<http://webbook.nist.gov/cgi/cbook.cgi?ID=C824555&Units=SI>

Joback Method:

https://en.wikipedia.org/wiki/Joback_method

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Crippen Method:

https://www.chemeo.com/doc/models/crippen_log10ws

Cheméo Relay (1.0):

Cheméo Relay (1.0, beta):

Legend

af:	Acentric Factor
cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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