

# 1-Phenanthrenecarboxaldehyde, 7-ethenyl-1,2,3,4,4a,4b,5,6,7,9,10,10a-dodecahydro- [1R-(1«alpha»,4a«beta»,4b«alpha»,7«beta»,10a«alpha»)]

Other names: Podocarp-8(14)-en-15-al, 13«alpha»-methyl-3-vinyl-  
Cryptopinon

Cryptopinone

Dextropimarinal

Pimaral

Pimarinal

**Inchi:** InChI=1S/C20H30O/c1-5-18(2)12-9-16-15(13-18)7-8-17-19(3,14-21)10-6-11-20(16,17)4/

**InchiKey:** JKBKXKTXDKYEOR-UHFFFAOYSA-N

**Formula:** C20H30O

**SMILES:** C=CC1(C)C=C2CCC3C(C)(C=O)CCCC3(C)C2CC1

**Mol. weight [g/mol]:** 286.45

**CAS:** 472-39-9

## Physical Properties

Property code	Value	Unit	Source
gf	216.03	kJ/mol	Joback Method
hf	-177.33	kJ/mol	Joback Method
hfus	16.55	kJ/mol	Joback Method
hvap	63.65	kJ/mol	Joback Method
log10ws	-5.66		Crippen Method
logp	5.321		Crippen Method
mcvol	253.050	ml/mol	McGowan Method
pc	1711.78	kPa	Joback Method
rinpol	2177.00		NIST Webbook
rinpol	2154.00		NIST Webbook
rinpol	2171.57		NIST Webbook
rinpol	2149.00		NIST Webbook
rinpol	2177.00		NIST Webbook
ripol	2761.00		NIST Webbook
ripol	2749.00		NIST Webbook
tb	739.43	K	Joback Method
tc	979.37	K	Joback Method
tf	468.12	K	Joback Method
vc	0.963	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	790.90	J/mol×K	739.43	Joback Method
cpg	816.13	J/mol×K	779.42	Joback Method
cpg	841.06	J/mol×K	819.41	Joback Method
cpg	866.16	J/mol×K	859.40	Joback Method
cpg	891.89	J/mol×K	899.39	Joback Method
cpg	918.71	J/mol×K	939.38	Joback Method
cpg	947.09	J/mol×K	979.37	Joback Method

## Sources

Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C472399&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C472399&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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