

# 2-Bromobenzoic acid, 3-methylbut-2-yl ester

<b>Inchi:</b>	InChI=1S/C12H15BrO2/c1-8(2)9(3)15-12(14)10-6-4-5-7-11(10)13/h4-9H,1-3H3
<b>InchiKey:</b>	FBOMHQXAOJVAFS-UHFFFAOYSA-N
<b>Formula:</b>	C12H15BrO2
<b>SMILES:</b>	CC(C)C(C)OC(=O)c1ccccc1Br
<b>Mol. weight [g/mol]:</b>	271.15

## Physical Properties

Property code	Value	Unit	Source
gf	-71.54	kJ/mol	Joback Method
hf	-294.98	kJ/mol	Joback Method
hfus	21.51	kJ/mol	Joback Method
hvap	60.06	kJ/mol	Joback Method
log10ws	-4.42		Crippen Method
logp	3.650		Crippen Method
mcvol	181.120	ml/mol	McGowan Method
pc	2704.22	kPa	Joback Method
rinpola	1678.00		NIST Webbook
tb	647.19	K	Joback Method
tc	875.08	K	Joback Method
tf	365.90	K	Joback Method
vc	0.673	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	434.91	J/molxK	647.19	Joback Method
cpg	449.28	J/molxK	685.17	Joback Method
cpg	462.69	J/molxK	723.15	Joback Method
cpg	475.16	J/molxK	761.14	Joback Method
cpg	486.75	J/molxK	799.12	Joback Method
cpg	497.48	J/molxK	837.10	Joback Method
cpg	507.38	J/molxK	875.08	Joback Method
dvisc	0.0019092	Paxs	365.90	Joback Method
dvisc	0.0009792	Paxs	412.78	Joback Method

dvisc	0.0005755	Paxs	459.66	Joback Method
dvisc	0.0003732	Paxs	506.55	Joback Method
dvisc	0.0002604	Paxs	553.43	Joback Method
dvisc	0.0001922	Paxs	600.31	Joback Method
dvisc	0.0001483	Paxs	647.19	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U354670&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U354670&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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