

# 6-chlorohexyl dichloroacetate

<b>Other names:</b>	1-Hexanol, 6-chloro, dichloroacetate
<b>Inchi:</b>	InChI=1S/C8H13Cl3O2/c9-5-3-1-2-4-6-13-8(12)7(10)11/h7H,1-6H2
<b>InchiKey:</b>	NTHUGYDCPBABME-UHFFFAOYSA-N
<b>Formula:</b>	C8H13Cl3O2
<b>SMILES:</b>	O=C(OCCCCCCI)C(Cl)Cl
<b>Mol. weight [g/mol]:</b>	247.55

## Physical Properties

Property code	Value	Unit	Source
gf	-255.67	kJ/mol	Joback Method
hf	-505.75	kJ/mol	Joback Method
hfus	28.33	kJ/mol	Joback Method
hvap	55.33	kJ/mol	Joback Method
log10ws	-3.10		Crippen Method
logp	3.133		Crippen Method
mcvol	167.740	ml/mol	McGowan Method
pc	2395.87	kPa	Joback Method
rinpol	1535.00		NIST Webbook
rinpol	1541.00		NIST Webbook
rinpol	1550.00		NIST Webbook
rinpol	1557.00		NIST Webbook
rinpol	1509.00		NIST Webbook
rinpol	1509.00		NIST Webbook
ripol	2281.00		NIST Webbook
ripol	2300.00		NIST Webbook
ripol	2312.00		NIST Webbook
ripol	2281.00		NIST Webbook
tb	570.58	K	Joback Method
tc	765.02	K	Joback Method
tf	326.84	K	Joback Method
vc	0.648	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	360.24	J/mol×K	570.58	Joback Method
cpg	410.08	J/mol×K	732.61	Joback Method
cpg	401.17	J/mol×K	700.21	Joback Method
cpg	391.74	J/mol×K	667.80	Joback Method
cpg	381.78	J/mol×K	635.39	Joback Method
cpg	371.28	J/mol×K	602.99	Joback Method
cpg	418.47	J/mol×K	765.02	Joback Method
dvisc	0.0002252	Paxs	570.58	Joback Method
dvisc	0.0002939	Paxs	529.96	Joback Method
dvisc	0.0004011	Paxs	489.33	Joback Method
dvisc	0.0005789	Paxs	448.71	Joback Method
dvisc	0.0008990	Paxs	408.09	Joback Method
dvisc	0.0015387	Paxs	367.46	Joback Method
dvisc	0.0030100	Paxs	326.84	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R112040&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R112040&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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