

# Malonic acid, 3,3-dimethylbut-2-yl propyl ester

<b>Inchi:</b>	InChI=1S/C12H22O4/c1-6-7-15-10(13)8-11(14)16-9(2)12(3,4)5/h9H,6-8H2,1-5H3
<b>InchiKey:</b>	AEURCQIDVNSBSL-UHFFFAOYSA-N
<b>Formula:</b>	C12H22O4
<b>SMILES:</b>	CCCOC(=O)CC(=O)OC(C)C(C)(C)C
<b>Mol. weight [g/mol]:</b>	230.30

## Physical Properties

Property code	Value	Unit	Source
gf	-417.28	kJ/mol	Joback Method
hf	-794.64	kJ/mol	Joback Method
hfus	21.47	kJ/mol	Joback Method
hvap	58.93	kJ/mol	Joback Method
log10ws	-2.44		Crippen Method
logp	2.308		Crippen Method
mcvol	194.820	ml/mol	McGowan Method
pc	1973.55	kPa	Joback Method
rinqol	1402.00		NIST Webbook
tb	622.87	K	Joback Method
tc	811.82	K	Joback Method
tf	356.74	K	Joback Method
vc	0.739	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	519.38	J/molxK	622.87	Joback Method
cpg	588.77	J/molxK	780.32	Joback Method
cpg	576.41	J/molxK	748.83	Joback Method
cpg	563.30	J/molxK	717.34	Joback Method
cpg	549.44	J/molxK	685.85	Joback Method
cpg	534.80	J/molxK	654.36	Joback Method
cpg	600.40	J/molxK	811.82	Joback Method
dvisc	0.0001246	Paxs	622.87	Joback Method
dvisc	0.0001685	Paxs	578.51	Joback Method

dvisc	0.0002397	Paxs	534.16	Joback Method
dvisc	0.0003635	Paxs	489.81	Joback Method
dvisc	0.0005987	Paxs	445.45	Joback Method
dvisc	0.0011014	Paxs	401.10	Joback Method
dvisc	0.0023577	Paxs	356.74	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U347201&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U347201&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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