

# Chlorbromuron

<b>Other names:</b>	1-(3-Chloro-4-bromophenyl)-3-methyl-3-methoxyurea 1-(4-Bromo-3-chlorophenyl)-3-methoxy-3-methylurea 3-(4-Bromo-3-Chlorophenyl)-1-methoxy-1-methylurea Bromex Bromex (herbicide) C 6313 CIBA 6313 Chlorobromuron Maloran N-(4-Bromo-3-chlorophenyl)-N'-methoxy-N'-methylurea Urea, 3-(4-bromo-3-chlorophenyl)-1-methoxy-1-methyl- Urea, N'-(4-bromo-3-chlorophenyl)-N-methoxy-N-methyl-
<b>Inchi:</b>	InChI=1S/C9H10BrClN2O2/c1-13(15-2)9(14)12-6-3-4-7(10)8(11)5-6/h3-5H,1-2H3,(H,12,
<b>InchiKey:</b>	NLYNUTMZTCLNOO-UHFFFAOYSA-N
<b>Formula:</b>	C9H10BrClN2O2
<b>SMILES:</b>	CON(C)C(=O)Nc1ccc(Br)c(Cl)c1
<b>Mol. weight [g/mol]:</b>	293.55
<b>CAS:</b>	13360-45-7

## Physical Properties

Property code	Value	Unit	Source
gf	86.69	kJ/mol	Joback Method
hf	-128.71	kJ/mol	Joback Method
hfus	32.72	kJ/mol	Joback Method
hvap	67.68	kJ/mol	Joback Method
log10ws	-3.92		Aqueous Solubility Prediction Method
log10ws	-3.92		Estimated Solubility Method
logp	3.128		Crippen Method
mvol	171.050	ml/mol	McGowan Method
pc	3493.01	kPa	Joback Method
tb	684.45	K	Joback Method
tc	914.71	K	Joback Method
tf	370.00 ± 0.20	K	NIST Webbook
tf	370.66 ± 0.20	K	NIST Webbook
vc	0.620	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	393.04	J/mol×K	684.45	Joback Method
cpg	403.69	J/mol×K	722.83	Joback Method
cpg	413.53	J/mol×K	761.20	Joback Method
cpg	422.60	J/mol×K	799.58	Joback Method
cpg	430.94	J/mol×K	837.96	Joback Method
cpg	438.58	J/mol×K	876.34	Joback Method
cpg	445.55	J/mol×K	914.71	Joback Method
hfust	26.54	kJ/mol	369.80	NIST Webbook
hfust	26.54	kJ/mol	369.80	NIST Webbook

## Sources

**Aqueous Solubility Prediction Method:** <http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa>

**Estimated Solubility Method:** [http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl\\_file/ci034243xsi20040112\\_053635.txt](http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt)

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C13360457&Units=SI>

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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