

# 1-Propylcyclopentene

<b>Other names:</b>	1-PROPYL-1-CYCLOPENTENE 1-n-Propylcyclopentene Cyclopentene, 1-propyl-
<b>Inchi:</b>	InChI=1S/C8H14/c1-2-5-8-6-3-4-7-8/h6H,2-5,7H2,1H3
<b>InchiKey:</b>	FLWGCAJANMGQBB-UHFFFAOYSA-N
<b>Formula:</b>	C8H14
<b>SMILES:</b>	CCCC1=CCCC1
<b>Mol. weight [g/mol]:</b>	110.20
<b>CAS:</b>	3074-61-1

## Physical Properties

Property code	Value	Unit	Source
gf	81.07	kJ/mol	Joback Method
hf	-81.32	kJ/mol	Joback Method
hfus	10.17	kJ/mol	Joback Method
hvap	34.92	kJ/mol	Joback Method
ie	8.48 ± 0.01	eV	NIST Webbook
log10ws	-2.92		Crippen Method
logp	2.897		Crippen Method
mcvol	108.420	ml/mol	McGowan Method
pc	3257.86	kPa	Joback Method
rinpol	839.20		NIST Webbook
rinpol	836.00		NIST Webbook
rinpol	839.00		NIST Webbook
rinpol	839.00		NIST Webbook
rinpol	836.00		NIST Webbook
rinpol	839.00		NIST Webbook
rinpol	838.00		NIST Webbook
rinpol	841.00		NIST Webbook
rinpol	840.00		NIST Webbook
rinpol	831.00		NIST Webbook
rinpol	841.00		NIST Webbook
rinpol	838.00		NIST Webbook
rinpol	862.00		NIST Webbook
rinpol	868.00		NIST Webbook
rinpol	836.50		NIST Webbook
rinpol	836.00		NIST Webbook

ripol	972.00			NIST Webbook
ripol	959.40			NIST Webbook
ripol	968.10			NIST Webbook
ripol	968.00			NIST Webbook
ripol	972.00			NIST Webbook
ripol	959.00			NIST Webbook
ripol	966.00			NIST Webbook
ripol	961.00			NIST Webbook
ripol	965.00			NIST Webbook
ripol	968.00			NIST Webbook
ripol	954.00			NIST Webbook
ripol	958.00			NIST Webbook
ripol	961.00			NIST Webbook
ripol	964.90			NIST Webbook
ripol	968.10			NIST Webbook
ripol	954.00			NIST Webbook
ripol	958.40			NIST Webbook
ripol	972.00			NIST Webbook
ripol	972.10			NIST Webbook
ripol	959.40			NIST Webbook
ripol	966.40			NIST Webbook
ripol	972.10			NIST Webbook
ripol	959.40			NIST Webbook
ripol	966.40			NIST Webbook
tb	405.20 ± 2.00		K	NIST Webbook
tb	405.00 ± 3.00		K	NIST Webbook
tb	452.00 ± 4.00		K	NIST Webbook
tc	603.30		K	Joback Method
tf	172.90 ± 0.50		K	NIST Webbook
tf	172.90 ± 2.00		K	NIST Webbook
vc	0.411		m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	202.93	J/molxK	406.53	Joback Method
cpg	217.41	J/molxK	439.32	Joback Method
cpg	231.16	J/molxK	472.12	Joback Method
cpg	244.21	J/molxK	504.91	Joback Method
cpg	256.57	J/molxK	537.71	Joback Method
cpg	268.28	J/molxK	570.50	Joback Method

cpg	279.37	J/molxK	603.30	Joback Method
dvisc	0.0035547	Paxs	208.34	Joback Method
dvisc	0.0017541	Paxs	241.37	Joback Method
dvisc	0.0010260	Paxs	274.40	Joback Method
dvisc	0.0006735	Paxs	307.43	Joback Method
dvisc	0.0004797	Paxs	340.47	Joback Method
dvisc	0.0003628	Paxs	373.50	Joback Method
dvisc	0.0002871	Paxs	406.53	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.40000e+01
Coeff. B	-3.24721e+03
Coeff. C	-5.90780e+01
Temperature range (K), min.	295.89
Temperature range (K), max.	432.81

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3074611&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3074611&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure: Crippen Method:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a> <a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.cheric.org/files/research/kdb/mol/mol635.mol">https://www.cheric.org/files/research/kdb/mol/mol635.mol</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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