

# 1,2-Propadiene-1,3-dione

<b>Other names:</b>	Carbon suboxide Carbon oxide (C3O2) C3O2
<b>Inchi:</b>	InChI=1S/C3O2/c4-2-1-3-5
<b>InchiKey:</b>	GNEVIACKFGQMHB-UHFFFAOYSA-N
<b>Formula:</b>	C3O2
<b>SMILES:</b>	O=C=C=C=O
<b>Mol. weight [g/mol]:</b>	68.03
<b>CAS:</b>	504-64-3

## Physical Properties

Property code	Value	Unit	Source
chl	-1059.10	kJ/mol	NIST Webbook
ea	0.85 ± 0.15	eV	NIST Webbook
gf	-37.68	kJ/mol	Joback Method
hf	-97.80 ± 1.80	kJ/mol	NIST Webbook
hfl	-121.50 ± 1.00	kJ/mol	NIST Webbook
hfus	20.53	kJ/mol	Joback Method
hvap	35.10	kJ/mol	Joback Method
ie	10.60	eV	NIST Webbook
ie	10.60 ± 0.03	eV	NIST Webbook
ie	10.61	eV	NIST Webbook
ie	10.60	eV	NIST Webbook
log10ws	-8.69		Crippen Method
logp	-0.639		Crippen Method
mcvol	47.670	ml/mol	McGowan Method
pc	9920.48	kPa	Joback Method
tb	255.45	K	Joback Method
tc	413.48	K	Joback Method
tf	180.00	K	Joback Method
vc	0.198	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	75.13	J/mol×K	255.45	Joback Method
cpg	76.52	J/mol×K	281.79	Joback Method
cpg	77.84	J/mol×K	308.13	Joback Method
cpg	79.11	J/mol×K	334.47	Joback Method
cpg	80.31	J/mol×K	360.81	Joback Method
cpg	81.46	J/mol×K	387.14	Joback Method
cpg	82.55	J/mol×K	413.48	Joback Method
hvapt	26.20	kJ/mol	205.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C504643&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C504643&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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