

# 3-acetyl-1,2-dithiolane

<b>Other names:</b>	1,2-Dithiolane, 3-acetyl
<b>Inchi:</b>	InChI=1S/C5H8OS2/c1-4(6)5-2-3-7-8-5/h5H,2-3H2,1H3
<b>InchiKey:</b>	OEBJHVCWENCTLE-UHFFFAOYSA-N
<b>Formula:</b>	C5H8OS2
<b>SMILES:</b>	CC(=O)C1CCSS1
<b>Mol. weight [g/mol]:</b>	148.25

## Physical Properties

Property code	Value	Unit	Source
gf	-21.43	kJ/mol	Joback Method
hf	-108.11	kJ/mol	Joback Method
hfus	11.55	kJ/mol	Joback Method
hvap	45.35	kJ/mol	Joback Method
log10ws	-1.96		Crippen Method
logp	1.729		Crippen Method
mcvol	104.720	ml/mol	McGowan Method
pc	4646.65	kPa	Joback Method
rinpol	1199.00		NIST Webbook
rinpol	1216.00		NIST Webbook
rinpol	1216.00		NIST Webbook
rinpol	1201.00		NIST Webbook
rinpol	1208.00		NIST Webbook
rinpol	1208.00		NIST Webbook
ripol	1994.00		NIST Webbook
ripol	1994.00		NIST Webbook
tb	478.61	K	Joback Method
tc	723.79	K	Joback Method
tf	373.84	K	Joback Method
vc	0.354	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	201.70	J/mol×K	478.61	Joback Method

cpg	213.54	J/mol×K	519.47	Joback Method
cpg	224.57	J/mol×K	560.34	Joback Method
cpg	234.82	J/mol×K	601.20	Joback Method
cpg	244.34	J/mol×K	642.06	Joback Method
cpg	253.15	J/mol×K	682.93	Joback Method
cpg	261.32	J/mol×K	723.79	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R125789&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R125789&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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