

# 1-Heptanethiol

<b>Other names:</b>	1-Heptylthiol HEPTYLTHIOL Heptyl mercaptan Heptyl thiol N-HEPTYL MERCAPTAN Normal-heptyl mercaptan USAF EK-2122 heptane-1-thiol n-Heptylmercaptan
<b>Inchi:</b>	InChI=1S/C7H16S/c1-2-3-4-5-6-7-8/h8H,2-7H2,1H3
<b>InchiKey:</b>	VPIAKHNXCOTPAY-UHFFFAOYSA-N
<b>Formula:</b>	C7H16S
<b>SMILES:</b>	CCCCCCS
<b>Mol. weight [g/mol]:</b>	132.27
<b>CAS:</b>	1639-09-4

## Physical Properties

Property code	Value	Unit	Source
chl	-5442.97 ± 0.75	kJ/mol	NIST Webbook
gf	37.45	kJ/mol	Joback Method
hf	-149.50 ± 0.96	kJ/mol	NIST Webbook
hfl	-200.50 ± 0.92	kJ/mol	NIST Webbook
hfus	17.93	kJ/mol	Joback Method
hvap	51.00	kJ/mol	NIST Webbook
hvap	50.60 ± 0.20	kJ/mol	NIST Webbook
log10ws	-2.82		Crippen Method
logp	2.887		Crippen Method
mcvol	125.840	ml/mol	McGowan Method
pc	2999.15	kPa	Joback Method
rinpol	1055.00		NIST Webbook
rinpol	1013.00		NIST Webbook
rinpol	1020.00		NIST Webbook
rinpol	1031.00		NIST Webbook
rinpol	1020.00		NIST Webbook
rinpol	1018.00		NIST Webbook
rinpol	1011.00		NIST Webbook
rinpol	1024.00		NIST Webbook

ripol	1024.00		NIST Webbook
ripol	1031.00		NIST Webbook
ripol	1031.00		NIST Webbook
ripol	1248.70		NIST Webbook
ripol	1267.00		NIST Webbook
ripol	1256.00		NIST Webbook
ripol	1253.80		NIST Webbook
ripol	1267.00		NIST Webbook
sl	375.35	J/molxK	NIST Webbook
tb	447.65	K	KDB
tb	450.20	K	NIST Webbook
tb	450.05 ± 0.30	K	NIST Webbook
tb	445.00 ± 2.00	K	NIST Webbook
tc	609.58	K	Joback Method
tf	229.80 ± 0.50	K	NIST Webbook
tf	229.80 ± 0.30	K	NIST Webbook
tt	229.93 ± 0.08	K	NIST Webbook
tt	229.92 ± 0.02	K	NIST Webbook
vc	0.481	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	312.38	J/molxK	609.58	Joback Method
cpg	291.46	J/molxK	547.19	Joback Method
cpg	302.15	J/molxK	578.38	Joback Method
cpg	243.63	J/molxK	422.42	Joback Method
cpg	256.37	J/molxK	453.61	Joback Method
cpg	268.58	J/molxK	484.81	Joback Method
cpg	280.27	J/molxK	516.00	Joback Method
cpl	259.32	J/molxK	298.15	NIST Webbook
cpl	259.32	J/molxK	298.15	NIST Webbook
hfust	25.40	kJ/mol	229.90	NIST Webbook
hfust	25.38	kJ/mol	229.92	NIST Webbook
hfust	25.40	kJ/mol	229.90	NIST Webbook
hvapt	45.00	kJ/mol	422.50	NIST Webbook
hvapt	49.50	kJ/mol	309.00	NIST Webbook
sfust	110.40	J/molxK	229.92	NIST Webbook

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	447.70	K	102.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.54403e+01
Coeff. B	-4.15483e+03
Coeff. C	-6.62750e+01
Temperature range (K), min.	340.47
Temperature range (K), max.	476.47

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	9.92087e+01
Coeff. B	-9.41115e+03
Coeff. C	-1.22765e+01
Coeff. D	6.51592e-06
Temperature range (K), min.	229.92
Temperature range (K), max.	645.00

## Sources

Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
KDB:	<a href="https://www.thermopedia.com/doc/thermophysical/kdb/mol/mol1840.mol">https://www.thermopedia.com/doc/thermophysical/kdb/mol/mol1840.mol</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1639094&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1639094&amp;Units=SI</a>
The Yaws Handbook of Vapor Pressure:	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
KDB Vapor Pressure Data:	<a href="https://www.thermopedia.com/doc/thermophysical/kdb/hcprop/showprop.php?cmpid=1840">https://www.thermopedia.com/doc/thermophysical/kdb/hcprop/showprop.php?cmpid=1840</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>sl:</b>	Liquid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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