

# Ethyl 3,5-dinitrobenzoate

<b>Other names:</b>	Benzoic acid, 3,5-dinitro-, ethyl ester
<b>Inchi:</b>	InChI=1S/C9H8N2O6/c1-2-17-9(12)6-3-7(10(13)14)5-8(4-6)11(15)16/h3-5H,2H2,1H3
<b>InchiKey:</b>	IBQREHJPMPCXQA-UHFFFAOYSA-N
<b>Formula:</b>	C9H8N2O6
<b>SMILES:</b>	CCOC(=O)c1cc([N+](=O)[O-])cc([N+](=O)[O-])c1
<b>Mol. weight [g/mol]:</b>	240.17
<b>CAS:</b>	618-71-3

## Physical Properties

Property code	Value	Unit	Source
gf	-44.77	kJ/mol	Joback Method
hf	-281.82	kJ/mol	Joback Method
hfus	37.84	kJ/mol	Joback Method
hvap	81.57	kJ/mol	Joback Method
log10ws	-3.43		Crippen Method
logp	1.680		Crippen Method
mcvol	156.190	ml/mol	McGowan Method
pc	3415.86	kPa	Joback Method
rinpol	1752.00		NIST Webbook
rinpol	1762.00		NIST Webbook
rinpol	1777.00		NIST Webbook
rinpol	1755.00		NIST Webbook
rinpol	1833.00		NIST Webbook
rinpol	1761.00		NIST Webbook
rinpol	1752.00		NIST Webbook
rinpol	1752.00		NIST Webbook
rinpol	1757.00		NIST Webbook
rinpol	1757.00		NIST Webbook
rinpol	1760.00		NIST Webbook
rinpol	1767.00		NIST Webbook
rinpol	1781.00		NIST Webbook
rinpol	1810.00		NIST Webbook
ripol	2739.00		NIST Webbook
ripol	2792.00		NIST Webbook
ripol	2739.00		NIST Webbook
ripol	2749.00		NIST Webbook
ripol	2753.00		NIST Webbook

ripol	2792.00		NIST Webbook
ripol	2774.00		NIST Webbook
ripol	2753.00		NIST Webbook
ripol	2781.00		NIST Webbook
ripol	2764.00		NIST Webbook
tb	821.93	K	Joback Method
tc	1082.48	K	Joback Method
tf	602.03	K	Joback Method
vc	0.620	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	427.66	J/mol×K	821.93	Joback Method
cpg	436.84	J/mol×K	865.36	Joback Method
cpg	445.02	J/mol×K	908.78	Joback Method
cpg	452.21	J/mol×K	952.21	Joback Method
cpg	458.45	J/mol×K	995.63	Joback Method
cpg	463.74	J/mol×K	1039.06	Joback Method
cpg	468.12	J/mol×K	1082.48	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C618713&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C618713&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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