

# Benzene, (1-propyldecyl)-

<b>Other names:</b>	Tridecane, 4-phenyl-
<b>Inchi:</b>	InChI=1S/C19H32/c1-3-5-6-7-8-9-11-15-18(14-4-2)19-16-12-10-13-17-19/h10,12-13,16-1
<b>InchiKey:</b>	RZGVZPAWCGDMCK-UHFFFAOYSA-N
<b>Formula:</b>	C19H32
<b>SMILES:</b>	CCCCCCCCC(CCC)c1ccccc1
<b>Mol. weight [g/mol]:</b>	260.46
<b>CAS:</b>	4534-51-4

## Physical Properties

Property code	Value	Unit	Source
gf	219.07	kJ/mol	Joback Method
hf	-204.24	kJ/mol	Joback Method
hfus	35.48	kJ/mol	Joback Method
hvap	59.78	kJ/mol	Joback Method
log10ws	-6.84		Crippen Method
logp	6.711		Crippen Method
mcvol	254.810	ml/mol	McGowan Method
pc	1368.70	kPa	Joback Method
rinpol	1838.00		NIST Webbook
rinpol	1838.00		NIST Webbook
rinpol	1833.00		NIST Webbook
ripol	2046.00		NIST Webbook
tb	660.36	K	Joback Method
tc	847.35	K	Joback Method
tf	315.31	K	Joback Method
vc	0.986	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	710.04	J/mol×K	660.36	Joback Method
cpg	730.40	J/mol×K	691.52	Joback Method
cpg	749.69	J/mol×K	722.69	Joback Method
cpg	767.96	J/mol×K	753.85	Joback Method

cpg	785.26	J/molxK	785.02	Joback Method
cpg	801.63	J/molxK	816.18	Joback Method
cpg	817.11	J/molxK	847.35	Joback Method
dvisc	0.0037196	Paxs	315.31	Joback Method
dvisc	0.0012862	Paxs	372.82	Joback Method
dvisc	0.0005907	Paxs	430.33	Joback Method
dvisc	0.0003259	Paxs	487.84	Joback Method
dvisc	0.0002039	Paxs	545.34	Joback Method
dvisc	0.0001395	Paxs	602.85	Joback Method
dvisc	0.0001019	Paxs	660.36	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4534514&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4534514&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/22-084-4/Benzene-1-propyldecyl.pdf>

Generated by Cheméo on 2024-04-27 22:48:09.087850616 +0000 UTC m=+16547338.008427960.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.