

# Heptyl caprylate

<b>Other names:</b>	Heptyl octanoate Octanoic acid, heptyl ester n-Heptyl n-octanoate
<b>Inchi:</b>	InChI=1S/C15H30O2/c1-3-5-7-9-11-13-15(16)17-14-12-10-8-6-4-2/h3-14H2,1-2H3
<b>InchiKey:</b>	TZXWLJYLYILFGM-UHFFFAOYSA-N
<b>Formula:</b>	C15H30O2
<b>SMILES:</b>	CCCCCCCCOC(=O)CCCCCCC
<b>Mol. weight [g/mol]:</b>	242.40
<b>CAS:</b>	4265-97-8

## Physical Properties

Property code	Value	Unit	Source
gf	-158.50	kJ/mol	Joback Method
hf	-597.73	kJ/mol	Joback Method
hfus	37.39	kJ/mol	Joback Method
hvap	58.14	kJ/mol	Joback Method
log10ws	-4.96		Crippen Method
logp	4.861		Crippen Method
mcvol	229.650	ml/mol	McGowan Method
pc	1465.73	kPa	Joback Method
ripol	1666.00		NIST Webbook
ripol	1666.00		NIST Webbook
ripol	1666.00		NIST Webbook
ripol	1892.00		NIST Webbook
ripol	1892.00		NIST Webbook
tb	572.00 ± 4.00	K	NIST Webbook
tb	563.80 ± 1.00	K	NIST Webbook
tc	786.41	K	Joback Method
tf	267.00 ± 2.00	K	NIST Webbook
tf	263.00 ± 1.00	K	NIST Webbook
vc	0.899	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	620.74	J/mol×K	618.89	Joback Method
cpg	638.14	J/mol×K	646.81	Joback Method
cpg	654.83	J/mol×K	674.73	Joback Method
cpg	670.81	J/mol×K	702.65	Joback Method
cpg	686.10	J/mol×K	730.57	Joback Method
cpg	700.70	J/mol×K	758.49	Joback Method
cpg	714.65	J/mol×K	786.41	Joback Method
dvisc	0.0025922	Paxs	330.97	Joback Method
dvisc	0.0011647	Paxs	378.96	Joback Method
dvisc	0.0006264	Paxs	426.94	Joback Method
dvisc	0.0003819	Paxs	474.93	Joback Method
dvisc	0.0002549	Paxs	522.92	Joback Method
dvisc	0.0001822	Paxs	570.90	Joback Method
dvisc	0.0001371	Paxs	618.89	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.51384e+01
Coeff. B	-4.90189e+03
Coeff. C	-9.78450e+01
Temperature range (K), min.	427.92
Temperature range (K), max.	596.67

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4265978&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4265978&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>ripolar:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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