

# 1,3-Dioxolane, 2-ethyl-4-((4-phenoxyphenoxy)methyl)-

**Other names:** 2-Ethyl-4-[(4-phenoxyphenoxy)methyl]-1,3-dioxolane

Diofenolan

**Inchi:** InChI=1S/C18H20O4/c1-2-18-20-13-17(22-18)12-19-14-8-10-16(11-9-14)21-15-6-4-3-5-7

**InchiKey:** ZDOOQPFIGYHZFV-UHFFFAOYSA-N

**Formula:** C18H20O4

**SMILES:** CCC1OCC(COc2ccc(Oc3ccccc3)cc2)O1

**Mol. weight [g/mol]:** 300.35

**CAS:** 63837-33-2

## Physical Properties

Property code	Value	Unit	Source
gf	-37.53	kJ/mol	Joback Method
hf	-441.56	kJ/mol	Joback Method
hfus	43.41	kJ/mol	Joback Method
hvap	74.66	kJ/mol	Joback Method
log10ws	-4.11		Crippen Method
logp	4.009		Crippen Method
mcvol	229.580	ml/mol	McGowan Method
pc	2058.62	kPa	Joback Method
rinpol	2459.00		NIST Webbook
rinpol	2459.00		NIST Webbook
tb	778.93	K	Joback Method
tc	1017.42	K	Joback Method
tf	462.24	K	Joback Method
vc	0.846	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	693.22	J/molxK	778.93	Joback Method
cpg	711.17	J/molxK	818.68	Joback Method
cpg	727.56	J/molxK	858.43	Joback Method
cpg	742.42	J/molxK	898.17	Joback Method
cpg	755.81	J/molxK	937.92	Joback Method

cpg	767.77	J/molxK	977.67	Joback Method
cpg	778.33	J/molxK	1017.42	Joback Method
dvisc	0.0008967	Paxs	462.24	Joback Method
dvisc	0.0005288	Paxs	515.02	Joback Method
dvisc	0.0003440	Paxs	567.80	Joback Method
dvisc	0.0002408	Paxs	620.59	Joback Method
dvisc	0.0001782	Paxs	673.37	Joback Method
dvisc	0.0001378	Paxs	726.15	Joback Method
dvisc	0.0001103	Paxs	778.93	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C63837332&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C63837332&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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