

# Butane, 1,1'-oxybis[4-chloro-

<b>Other names:</b>	Ether, bis(4-chlorobutyl) Bis(4-chlorobutyl) ether Di(chlorobutyl) ether 4,4'-Dichlorodibutyl ether 4-Chlorobutyl ether 4-Chloro-n-butyl ether Bis-(4-chlorbut-1-yl)-ether 1,1'-Oxydi-4-chlorobutane 1,1'-Oxybis(4-chlorobutane) NSC 26978
<b>Inchi:</b>	InChI=1S/C8H16Cl2O/c9-5-1-3-7-11-8-4-2-6-10/h1-8H2
<b>InchiKey:</b>	PVBMXMKIKMJQRK-UHFFFAOYSA-N
<b>Formula:</b>	C8H16Cl2O
<b>SMILES:</b>	C1CCCCOCCCC1
<b>Mol. weight [g/mol]:</b>	199.12
<b>CAS:</b>	6334-96-9

## Physical Properties

Property code	Value	Unit	Source
gf	-112.38	kJ/mol	Joback Method
hf	-372.15	kJ/mol	Joback Method
hfus	26.06	kJ/mol	Joback Method
hvap	44.58	kJ/mol	Joback Method
log10ws	-2.56		Crippen Method
logp	3.041		Crippen Method
mcvol	153.930	ml/mol	McGowan Method
pc	2318.07	kPa	Joback Method
tb	400.00 ± 4.00	K	NIST Webbook
tb	400.00 ± 4.00	K	NIST Webbook
tc	656.55	K	Joback Method
tf	261.99	K	Joback Method
vc	0.600	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	316.20	J/molxK	479.72	Joback Method
cpg	328.33	J/molxK	509.19	Joback Method
cpg	339.99	J/molxK	538.66	Joback Method
cpg	351.18	J/molxK	568.14	Joback Method
cpg	361.93	J/molxK	597.61	Joback Method
cpg	372.23	J/molxK	627.08	Joback Method
cpg	382.09	J/molxK	656.55	Joback Method
dvisc	0.0034133	Paxs	261.99	Joback Method
dvisc	0.0016785	Paxs	298.28	Joback Method
dvisc	0.0009628	Paxs	334.57	Joback Method
dvisc	0.0006158	Paxs	370.86	Joback Method
dvisc	0.0004265	Paxs	407.14	Joback Method
dvisc	0.0003137	Paxs	443.43	Joback Method
dvisc	0.0002417	Paxs	479.72	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6334969&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6334969&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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