

# Glutaric acid, 2-(2-chlorophenoxy)ethyl isobutyl ester

<b>Inchi:</b>	InChI=1S/C17H23ClO5/c1-13(2)12-23-17(20)9-5-8-16(19)22-11-10-21-15-7-4-3-6-14(15)
<b>InchiKey:</b>	BUCPPJVHXUXFLX-UHFFFAOYSA-N
<b>Formula:</b>	C17H23ClO5
<b>SMILES:</b>	CC(C)COC(=O)CCCC(=O)OCCOc1ccccc1Cl
<b>Mol. weight [g/mol]:</b>	342.81

## Physical Properties

Property code	Value	Unit	Source
gf	-392.17	kJ/mol	Joback Method
hf	-811.99	kJ/mol	Joback Method
hfus	40.87	kJ/mol	Joback Method
hvap	81.09	kJ/mol	Joback Method
log10ws	-3.95		Crippen Method
logp	3.631		Crippen Method
mcvol	259.620	ml/mol	McGowan Method
pc	1615.47	kPa	Joback Method
rinpol	2477.00		NIST Webbook
rinpol	2477.00		NIST Webbook
tb	832.01	K	Joback Method
tc	1038.87	K	Joback Method
tf	501.76	K	Joback Method
vc	0.989	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	759.60	J/molxK	832.01	Joback Method
cpg	818.31	J/molxK	1004.39	Joback Method
cpg	808.79	J/molxK	969.92	Joback Method
cpg	798.17	J/molxK	935.44	Joback Method
cpg	786.43	J/molxK	900.96	Joback Method
cpg	773.58	J/molxK	866.49	Joback Method
cpg	826.72	J/molxK	1038.87	Joback Method
dvisc	0.0000498	Paxs	832.01	Joback Method

dvisc	0.0000641	Paxs	776.97	Joback Method
dvisc	0.0000857	Paxs	721.93	Joback Method
dvisc	0.0001203	Paxs	666.88	Joback Method
dvisc	0.0001794	Paxs	611.84	Joback Method
dvisc	0.0002894	Paxs	556.80	Joback Method
dvisc	0.0005186	Paxs	501.76	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U377316&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U377316&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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