

# 1,2,3-Benzenetricarboxylic acid

<b>Other names:</b>	Benzene-1,2,3-tricarboxylic acid Hemimellitic acid 1,2,3-Tricarboxybenzene
<b>Inchi:</b>	InChI=1S/C9H6O6/c10-7(11)4-2-1-3-5(8(12)13)6(4)9(14)15/h1-3H,(H,10,11)(H,12,13)(H,13)
<b>InchiKey:</b>	UJMDYLWCYJJYMO-UHFFFAOYSA-N
<b>Formula:</b>	C9H6O6
<b>SMILES:</b>	O=C(O)c1cccc(C(=O)O)c1C(=O)O
<b>Mol. weight [g/mol]:</b>	210.14
<b>CAS:</b>	569-51-7

## Physical Properties

Property code	Value	Unit	Source
chs	-3238.70 ± 0.84	kJ/mol	NIST Webbook
gf	-679.17	kJ/mol	Joback Method
hf	-809.93	kJ/mol	Joback Method
hfs	-1160.40 ± 0.84	kJ/mol	NIST Webbook
hfus	29.39	kJ/mol	Joback Method
hvap	109.50	kJ/mol	Joback Method
log10ws	-1.65		Crippen Method
logp	0.781		Crippen Method
mcvol	136.230	ml/mol	McGowan Method
pc	5774.15	kPa	Joback Method
tb	880.11	K	Joback Method
tc	1085.73	K	Joback Method
tf	372.00 ± 3.00	K	NIST Webbook
tf	482.00 ± 2.00	K	NIST Webbook
vc	0.506	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	366.92	J/molxK	880.11	Joback Method
cpg	371.82	J/molxK	914.38	Joback Method
cpg	376.27	J/molxK	948.65	Joback Method

cpg	380.27	J/mol×K	982.92	Joback Method
cpg	383.85	J/mol×K	1017.19	Joback Method
cpg	387.03	J/mol×K	1051.46	Joback Method
cpg	389.82	J/mol×K	1085.73	Joback Method
dvisc	0.0000920	Paxs	574.90	Joback Method
dvisc	0.0000343	Paxs	625.77	Joback Method
dvisc	0.0000148	Paxs	676.64	Joback Method
dvisc	0.0000072	Paxs	727.50	Joback Method
dvisc	0.0000038	Paxs	778.37	Joback Method
dvisc	0.0000022	Paxs	829.24	Joback Method
dvisc	0.0000014	Paxs	880.11	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C569517&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C569517&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/22-311-1/1-2-3-Benzenetricarboxylic-acid.pdf>

Generated by Cheméo on 2024-04-26 20:10:32.436056915 +0000 UTC m=+16451481.356634237.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.