

# tungsten

Other names:	wolfram
Inchi:	InChI=1S/W
InchiKey:	WFKWXMTUELFFGS-UHFFFAOYSA-N
Formula:	W
SMILES:	[W]
Mol. weight [g/mol]:	183.84
CAS:	7440-33-7

## Physical Properties

Property code	Value	Unit	Source
ea	0.82 ± 0.00	eV	NIST Webbook
ea	0.82 ± 0.00	eV	NIST Webbook
ea	0.82 ± 0.01	eV	NIST Webbook
ie	7.98	eV	NIST Webbook
ie	7.98	eV	NIST Webbook
ie	7.49 ± 0.08	eV	NIST Webbook
ie	7.98	eV	NIST Webbook
ie	7.98	eV	NIST Webbook

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
dvisc	0.0167000	Paxs	3155.00	Viscosity of molten Mo, Ta, Os, Re, and W measured by electrostatic levitation
dvisc	0.0157000	Paxs	3200.00	Viscosity of molten Mo, Ta, Os, Re, and W measured by electrostatic levitation

dvisc	0.0146000	Paxs	3250.00	Viscosity of molten Mo, Ta, Os, Re, and W measured by electrostatic levitation
dvisc	0.0137000	Paxs	3300.00	Viscosity of molten Mo, Ta, Os, Re, and W measured by electrostatic levitation
dvisc	0.0128000	Paxs	3350.00	Viscosity of molten Mo, Ta, Os, Re, and W measured by electrostatic levitation
dvisc	0.0120000	Paxs	3400.00	Viscosity of molten Mo, Ta, Os, Re, and W measured by electrostatic levitation
dvisc	0.0113000	Paxs	3450.00	Viscosity of molten Mo, Ta, Os, Re, and W measured by electrostatic levitation
dvisc	0.0106000	Paxs	3500.00	Viscosity of molten Mo, Ta, Os, Re, and W measured by electrostatic levitation
dvisc	0.0100000	Paxs	3550.00	Viscosity of molten Mo, Ta, Os, Re, and W measured by electrostatic levitation
dvisc	0.0094000	Paxs	3600.00	Viscosity of molten Mo, Ta, Os, Re, and W measured by electrostatic levitation
dvisc	0.0091000	Paxs	3634.00	Viscosity of molten Mo, Ta, Os, Re, and W measured by electrostatic levitation
dvisc	0.0085000	Paxs	3695.00	Viscosity of molten Mo, Ta, Os, Re, and W measured by electrostatic levitation

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	2.09719e+01
Coeff. B	-9.29967e+04
Coeff. C	-1.41510e+02
Temperature range (K), min.	3477.15
Temperature range (K), max.	5828.15

## Sources

NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7440337&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7440337&amp;Units=SI</a>
The Yaws Handbook of Vapor Pressure:	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
A calorimetric and thermodynamic investigation of potassium uranyl manganate, $K_2MnO_4 \cdot 2H_2O$ , Os, Re, and W measured by electrostatic levitation:	<a href="https://www.doi.org/10.1016/j.jct.2012.09.014">https://www.doi.org/10.1016/j.jct.2012.09.014</a>
A calorimetric investigation of $A_2[(UO_2)_2(WO_5)_2O]$ compounds with A = K, Rb and Cs and the associated phase transition of $Cs_2[(UO_2)_2(WO_5)_2O]$ and $A_2[(UO_2)_2(WO_5)_2O]$ systems:	<a href="https://www.doi.org/10.1016/j.jct.2013.05.036">https://www.doi.org/10.1016/j.jct.2013.05.036</a>
A calorimetric and thermodynamic investigation of $Cs_2[(UO_2)_2(WO_5)_2O]$ and $A_2[(UO_2)_2(WO_5)_2O]$ systems:	<a href="https://www.doi.org/10.1016/j.jct.2017.03.039">https://www.doi.org/10.1016/j.jct.2017.03.039</a>
A study of the thermodynamic properties of the $Cs_2[(UO_2)_2(WO_5)_2O]$ and $A_2[(UO_2)_2(WO_5)_2O]$ systems:	<a href="https://www.doi.org/10.1016/j.jct.2019.05.012">https://www.doi.org/10.1016/j.jct.2019.05.012</a>
Thermal conductivity of the $Cs_2[(UO_2)_2(WO_5)_2O]$ and $A_2[(UO_2)_2(WO_5)_2O]$ systems:	<a href="https://www.doi.org/10.1016/j.jct.2019.07.015">https://www.doi.org/10.1016/j.jct.2019.07.015</a>
Thermal conductivity of the $Cs_2[(UO_2)_2(WO_5)_2O]$ and $A_2[(UO_2)_2(WO_5)_2O]$ systems:	<a href="https://www.doi.org/10.1016/j.tca.2012.03.022">https://www.doi.org/10.1016/j.tca.2012.03.022</a>

## Legend

<b>dvisc:</b>	Dynamic viscosity
<b>ea:</b>	Electron affinity
<b>ie:</b>	Ionization energy
<b>pvap:</b>	Vapor pressure

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