

# Tricyclo[4.2.1.0<sup>2,5</sup>]non-7-ene,(1 «alpha»,2 «alpha»,

<b>Inchi:</b>	InChI=1S/C9H12/c1-2-7-5-6(1)8-3-4-9(7)8/h3-4,6-9H,1-2,5H2/t6-,7-,8-,9+/m0/s1
<b>InchiKey:</b>	HEYDBCUMMFxDBQ-XSPKLOCKSA-N
<b>Formula:</b>	C9H12
<b>SMILES:</b>	C1=CC2C3CCC(C3)C12
<b>Mol. weight [g/mol]:</b>	120.19
<b>CAS:</b>	16529-83-2

## Physical Properties

Property code	Value	Unit	Source
gf	229.40	kJ/mol	Joback Method
hf	26.75	kJ/mol	Joback Method
hfus	15.76	kJ/mol	Joback Method
hvap	35.35	kJ/mol	Joback Method
ie	8.92 ± 0.03	eV	NIST Webbook
log10ws	-2.16		Crippen Method
logp	2.219		Crippen Method
mcvol	100.790	ml/mol	McGowan Method
pc	3505.43	kPa	Joback Method
tb	420.03	K	Joback Method
tc	629.91	K	Joback Method
tf	241.53	K	Joback Method
vc	0.396	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	214.61	J/mol×K	420.03	Joback Method
cpg	292.17	J/mol×K	594.93	Joback Method
cpg	278.99	J/mol×K	559.95	Joback Method
cpg	264.74	J/mol×K	524.97	Joback Method
cpg	249.32	J/mol×K	489.99	Joback Method
cpg	232.64	J/mol×K	455.01	Joback Method
cpg	304.38	J/mol×K	629.91	Joback Method
dvisc	0.0009565	Paxs	420.03	Joback Method

dvisc	0.0008374	Paxs	390.28	Joback Method
dvisc	0.0007171	Paxs	360.53	Joback Method
dvisc	0.0005973	Paxs	330.78	Joback Method
dvisc	0.0004798	Paxs	301.03	Joback Method
dvisc	0.0003673	Paxs	271.28	Joback Method
dvisc	0.0002633	Paxs	241.53	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C16529832&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C16529832&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/22-343-6/Tricyclo-4-2-1-02-5-non-7-ene-1-alpha-2-alpha-5-alpha-6-alpha.pdf>

Generated by Cheméo on 2024-04-19 21:34:15.419817564 +0000 UTC m=+15851704.340394876.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.