

# Fumaric acid, 3-methylbut-2-yl octadecyl ester

<b>Inchi:</b>	InChI=1S/C27H50O4/c1-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-23-30-26(28)21-2
<b>InchiKey:</b>	ZFDMIYUQICYKLT-QURGRASLSA-N
<b>Formula:</b>	C27H50O4
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCOC(=O)C=CC(=O)OC(C)C(C)C
<b>Mol. weight [g/mol]:</b>	438.68

## Physical Properties

Property code	Value	Unit	Source
gf	-216.04	kJ/mol	Joback Method
hf	-983.55	kJ/mol	Joback Method
hfus	64.42	kJ/mol	Joback Method
hvap	93.19	kJ/mol	Joback Method
log10ws	-8.57		Crippen Method
logp	7.935		Crippen Method
mvol	401.870	ml/mol	McGowan Method
pc	750.61	kPa	Joback Method
rinpol	2955.00		NIST Webbook
rinpol	2955.00		NIST Webbook
tb	973.02	K	Joback Method
tc	1197.14	K	Joback Method
tf	503.29	K	Joback Method
vc	1.563	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1382.46	J/molxK	973.02	Joback Method
cpg	1403.49	J/molxK	1010.37	Joback Method
cpg	1422.90	J/molxK	1047.73	Joback Method
cpg	1440.75	J/molxK	1085.08	Joback Method
cpg	1457.13	J/molxK	1122.43	Joback Method
cpg	1472.09	J/molxK	1159.79	Joback Method
cpg	1485.71	J/molxK	1197.14	Joback Method
dvisc	0.0004419	Paxs	503.29	Joback Method

dvisc	0.0001669	Paxs	581.58	Joback Method
dvisc	0.0000794	Paxs	659.87	Joback Method
dvisc	0.0000442	Paxs	738.15	Joback Method
dvisc	0.0000276	Paxs	816.44	Joback Method
dvisc	0.0000187	Paxs	894.73	Joback Method
dvisc	0.0000134	Paxs	973.02	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348089&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348089&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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