

# Propanoic acid, 2-mercapto-, methyl ester

<b>Other names:</b>	Methyl 2-mercaptopropionate Methyl «alpha»-mercaptopropionate «alpha»-Mercaptopropionic acid methyl ester Methyl alpha-mercaptopropionate
<b>Inchi:</b>	InChI=1S/C4H8O2S/c1-3(7)4(5)6-2/h3,7H,1-2H3
<b>InchiKey:</b>	SNWKNPMDQONHKK-UHFFFAOYSA-N
<b>Formula:</b>	C4H8O2S
<b>SMILES:</b>	COC(=O)C(C)S
<b>Mol. weight [g/mol]:</b>	120.17
<b>CAS:</b>	53907-46-3

## Physical Properties

Property code	Value	Unit	Source
gf	-224.17	kJ/mol	Joback Method
hf	-337.49	kJ/mol	Joback Method
hfus	9.42	kJ/mol	Joback Method
hvap	40.00	kJ/mol	Joback Method
log10ws	-0.54		Crippen Method
logp	0.478		Crippen Method
mcvol	91.010	ml/mol	McGowan Method
pc	4540.80	kPa	Joback Method
rinpol	813.00		NIST Webbook
tb	429.63	K	Joback Method
tc	638.94	K	Joback Method
tf	228.46	K	Joback Method
vc	0.332	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	166.98	J/molxK	429.63	Joback Method
cpg	175.08	J/molxK	464.52	Joback Method
cpg	182.89	J/molxK	499.40	Joback Method
cpg	190.40	J/molxK	534.29	Joback Method

cpg	197.60	J/mol×K	569.17	Joback Method
cpg	204.49	J/mol×K	604.06	Joback Method
cpg	211.07	J/mol×K	638.94	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C53907463&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C53907463&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/22-506-5/Propanoic-acid-2-mercapto-methyl-ester.pdf>

Generated by Cheméo on 2024-04-29 13:15:37.565291767 +0000 UTC m=+16685786.485869079.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.