

# Octanoic acid, 2-ethylhexyl ester

|                      |                                                                                  |
|----------------------|----------------------------------------------------------------------------------|
| Other names:         | 2-Ethylhexyl caprylate<br>2-ethylhexyl octanoate                                 |
| Inchi:               | InChI=1S/C16H32O2/c1-4-7-9-10-11-13-16(17)18-14-15(6-3)12-8-5-2/h15H,4-14H2,1-3H |
| InchiKey:            | HPUAIVNIHNEYPO-UHFFFAOYSA-N                                                      |
| Formula:             | C16H32O2                                                                         |
| SMILES:              | CCCCCCCC(=O)OCC(CC)CCCC                                                          |
| Mol. weight [g/mol]: | 256.42                                                                           |
| CAS:                 | 63321-70-0                                                                       |

## Physical Properties

| Property code | Value   | Unit    | Source         |
|---------------|---------|---------|----------------|
| gf            | -152.52 | kJ/mol  | Joback Method  |
| hf            | -623.65 | kJ/mol  | Joback Method  |
| hfus          | 36.46   | kJ/mol  | Joback Method  |
| hvap          | 59.98   | kJ/mol  | Joback Method  |
| log10ws       | -5.14   |         | Crippen Method |
| logp          | 5.107   |         | Crippen Method |
| mcvol         | 243.740 | ml/mol  | McGowan Method |
| pc            | 1371.74 | kPa     | Joback Method  |
| rinpol        | 1688.00 |         | NIST Webbook   |
| rinpol        | 1688.00 |         | NIST Webbook   |
| tb            | 641.33  | K       | Joback Method  |
| tc            | 810.93  | K       | Joback Method  |
| tf            | 327.24  | K       | Joback Method  |
| vc            | 0.950   | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 676.23 | J/mol×K | 641.33          | Joback Method |
| cpg           | 759.39 | J/mol×K | 782.67          | Joback Method |
| cpg           | 744.26 | J/mol×K | 754.40          | Joback Method |
| cpg           | 728.40 | J/mol×K | 726.13          | Joback Method |
| cpg           | 711.78 | J/mol×K | 697.86          | Joback Method |

|       |           |         |        |               |
|-------|-----------|---------|--------|---------------|
| cpg   | 694.40    | J/mol×K | 669.60 | Joback Method |
| cpg   | 773.80    | J/mol×K | 810.93 | Joback Method |
| dvisc | 0.0001135 | Pa×s    | 641.33 | Joback Method |
| dvisc | 0.0001544 | Pa×s    | 588.98 | Joback Method |
| dvisc | 0.0002232 | Pa×s    | 536.63 | Joback Method |
| dvisc | 0.0003493 | Pa×s    | 484.28 | Joback Method |
| dvisc | 0.0006092 | Pa×s    | 431.94 | Joback Method |
| dvisc | 0.0012389 | Pa×s    | 379.59 | Joback Method |
| dvisc | 0.0031617 | Pa×s    | 327.24 | Joback Method |

## Sources

|                        |                                                                                                                                               |
|------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------|
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                         |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                         |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C63321700&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C63321700&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                     |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                             |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>rinpol:</b>  | Non-polar retention indices                     |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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