

# 2-Propenoic acid, tetradecyl ester

<b>Other names:</b>	2-Propenoic acid, n-tetradecyl ester Tetradecyl acrylate
<b>Inchi:</b>	InChI=1S/C17H32O2/c1-3-5-6-7-8-9-10-11-12-13-14-15-16-19-17(18)4-2/h4H,2-3,5-16H
<b>InchiKey:</b>	XZHNPVKXBNJDGJD-UHFFFAOYSA-N
<b>Formula:</b>	C17H32O2
<b>SMILES:</b>	C=CC(=O)OCCCCCCCCCCCCCCC
<b>Mol. weight [g/mol]:</b>	268.43
<b>CAS:</b>	21643-42-5

## Physical Properties

Property code	Value	Unit	Source
gf	-53.82	kJ/mol	Joback Method
hf	-513.58	kJ/mol	Joback Method
hfus	41.29	kJ/mol	Joback Method
hvap	61.92	kJ/mol	Joback Method
log10ws	-5.65		Crippen Method
logp	5.417		Crippen Method
mcvol	253.530	ml/mol	McGowan Method
pc	1310.85	kPa	Joback Method
rinpol	1876.00		NIST Webbook
ripol	2201.00		NIST Webbook
ripol	2201.00		NIST Webbook
tb	661.33	K	Joback Method
tc	830.65	K	Joback Method
tf	287.00 ± 3.00	K	NIST Webbook
vc	0.993	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	804.54	J/mol×K	830.65	Joback Method
cpg	790.43	J/mol×K	802.43	Joback Method
cpg	775.61	J/mol×K	774.21	Joback Method
cpg	760.05	J/mol×K	745.99	Joback Method

cpg	743.76	J/mol×K	717.77	Joback Method
cpg	726.69	J/mol×K	689.55	Joback Method
cpg	708.83	J/mol×K	661.33	Joback Method
dvisc	0.0021607	Paxs	351.75	Joback Method
dvisc	0.0001135	Paxs	661.33	Joback Method
dvisc	0.0001506	Paxs	609.73	Joback Method
dvisc	0.0002107	Paxs	558.14	Joback Method
dvisc	0.0003156	Paxs	506.54	Joback Method
dvisc	0.0005182	Paxs	454.94	Joback Method
dvisc	0.0009658	Paxs	403.35	Joback Method
hvapt	69.40	kJ/mol	529.50	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C21643425&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C21643425&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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