Phenol, 3-phenoxy-

Other names:	3-Hydroxydiphenyl ether
	Phenol, m-phenoxy-
	m-Phenoxyphenol
	3-Phenoxyphenol
Inchi:	InChI=1S/C12H10O2/c13-10-5-4-8-12(9-10)14-11-6-2-1-3-7-11/h1-9,13H
InchiKey:	HBUCPZGYBSEEHF-UHFFFAOYSA-N
Formula:	C12H10O2
SMILES:	Oc1cccc(Oc2ccccc2)c1
Mol. weight [g/mol]:	186.21
CAS:	713-68-8

Physical Properties

Property code	Value	Unit	Source
gf	15.36	kJ/mol	Joback Method
hf	-127.48	kJ/mol	Joback Method
hfus	21.89	kJ/mol	Joback Method
hvap	62.28	kJ/mol	Joback Method
log10ws	-2.65		Crippen Method
logp	3.184		Crippen Method
mcvol	144.160	ml/mol	McGowan Method
pc	4026.13	kPa	Joback Method
rinpol	1731.00		NIST Webbook
tb	630.36	K	Joback Method
tc	886.36	K	Joback Method
tf	411.79	K	Joback Method
vc	0.475	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	352.76	J/mol×K	630.36	Joback Method
cpg	366.38	J/mol×K	673.03	Joback Method
cpg	378.88	J/mol×K	715.69	Joback Method
cpg	390.37	J/mol×K	758.36	Joback Method

cpg	400.97	J/mol×K	801.02	Joback Method
cpg	410.81	J/mol×K	843.69	Joback Method
cpg	420.01	J/mol×K	886.36	Joback Method
dvisc	0.0007478	Pa×s	411.79	Joback Method
dvisc	0.0003302	Pa×s	448.22	Joback Method
dvisc	0.0001648	Pa×s	484.65	Joback Method
dvisc	0.0000907	Pa×s	521.08	Joback Method
dvisc	0.0000540	Pa×s	557.50	Joback Method
dvisc	0.0000342	Pa×s	593.93	Joback Method
dvisc	0.0000229	Pa×s	630.36	Joback Method
hvapt	69.50	kJ/mol	455.00	NIST Webbook

Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	458.20	K	1.60	NIST Webbook

Sources

Crippen Method: https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method: https://en.wikipedia.org/wiki/Joback_method
McGowan Method: http://link.springer.com/article/10.1007/BF02311772
NIST Webbook: http://webbook.nist.gov/cgi/cbook.cgi?ID=C713688&Units=SI
Crippen Method: http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg: Ideal gas heat capacity
dvisc: Dynamic viscosity
gf: Standard Gibbs free energy of formation
hf: Enthalpy of formation at standard conditions
hfus: Enthalpy of fusion at standard conditions
hvap: Enthalpy of vaporization at standard conditions
hvapt: Enthalpy of vaporization at a given temperature
log10ws: Log10 of Water solubility in mol/l

logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tbrp:	Boiling point at reduced pressure
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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