

# Benzene, 1-methyl-4-(2-methylpropyl)-

<b>Other names:</b>	Toluene, p-isobutyl-
<b>Inchi:</b>	InChI=1S/C11H16/c1-9(2)8-11-6-4-10(3)5-7-11/h4-7,9H,8H2,1-3H3
<b>InchiKey:</b>	VCGBZXLLPCGFQM-UHFFFAOYSA-N
<b>Formula:</b>	C11H16
<b>SMILES:</b>	<chem>Cc1ccc(CC(C)C)cc1</chem>
<b>Mol. weight [g/mol]:</b>	148.24
<b>CAS:</b>	5161-04-6

## Physical Properties

Property code	Value	Unit	Source
gf	142.08	kJ/mol	Joback Method
hf	-50.59	kJ/mol	Joback Method
hfus	14.37	kJ/mol	Joback Method
hvap	42.63	kJ/mol	Joback Method
log10ws	-3.35		Crippen Method
logp	3.194		Crippen Method
mcvol	142.090	ml/mol	McGowan Method
pc	2629.85	kPa	Joback Method
tb	470.00 ± 6.00	K	NIST Webbook
tc	690.17	K	Joback Method
tf	237.67	K	Joback Method
vc	0.537	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	300.93	J/mol×K	482.30	Joback Method
cpg	373.16	J/mol×K	655.52	Joback Method
cpg	360.30	J/mol×K	620.88	Joback Method
cpg	346.68	J/mol×K	586.23	Joback Method
cpg	332.27	J/mol×K	551.59	Joback Method
cpg	317.03	J/mol×K	516.94	Joback Method
cpg	385.28	J/mol×K	690.17	Joback Method
dvisc	0.0001987	Paxs	482.30	Joback Method

dvisc	0.0002582	Paxs	441.53	Joback Method
dvisc	0.0003537	Paxs	400.76	Joback Method
dvisc	0.0005204	Paxs	359.99	Joback Method
dvisc	0.0008449	Paxs	319.21	Joback Method
dvisc	0.0015812	Paxs	278.44	Joback Method
dvisc	0.0036688	Paxs	237.67	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5161046&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5161046&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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