

# 2-[2-[2-(2-Methoxyethoxy)ethoxy]ethoxy]ethyl 2,2,3,3,3-pentafluoropropanoate

Other names:

Tetraethylene glycol monomethyl ether, pentafluoropropionate

3,6,9,12-Tetraoxatridec-1-yl pentafluoropropionate

Inchi:

InChI=1S/C12H19F5O6/c1-19-2-3-20-4-5-21-6-7-22-8-9-23-10(18)11(13,14)12(15,16)17

InchiKey:

CIPYBQIMJRGDJD-UHFFFAOYSA-N

Formula:

C12H19F5O6

SMILES:

COCCOCCOCCOCCOC(=O)C(F)(F)C(F)(F)F

Mol. weight [g/mol]:

354.27

## Physical Properties

Property code	Value	Unit	Source
gf	-1572.13	kJ/mol	Joback Method
hf	-2062.74	kJ/mol	Joback Method
hfus	34.95	kJ/mol	Joback Method
hvap	54.42	kJ/mol	Joback Method
log10ws	-1.03		Crippen Method
logp	1.423		Crippen Method
mcvol	219.710	ml/mol	McGowan Method
pc	1503.48	kPa	Joback Method
rinpol	1484.80		NIST Webbook
tb	629.82	K	Joback Method
tc	786.43	K	Joback Method
tf	393.87	K	Joback Method
vc	0.872	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	622.04	J/molxK	629.82	Joback Method
cpg	635.54	J/molxK	655.92	Joback Method
cpg	648.45	J/molxK	682.02	Joback Method
cpg	660.75	J/molxK	708.13	Joback Method
cpg	672.46	J/molxK	734.23	Joback Method
cpg	683.57	J/molxK	760.33	Joback Method
cpg	694.10	J/molxK	786.43	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U351970&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U351970&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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