

# N,N-Dimethyl-N'-octyl-p-methoxybenzamidine

Inchi:	InChI=1S/C18H30N2O/c1-5-6-7-8-9-10-15-19-18(20(2)3)16-11-13-17(21-4)14-12-16/h11
InchiKey:	AKQHCILIWABBCM-VHEBQXMUSA-N
Formula:	C18H30N2O
SMILES:	CCCCCCCCN=C(c1ccc(OC)cc1)N(C)C
Mol. weight [g/mol]:	290.44

## Physical Properties

Property code	Value	Unit	Source
hf	-182.05	kJ/mol	Joback Method
hvap	66.45	kJ/mol	Joback Method
log10ws	-4.49		Crippen Method
logp	4.364		Crippen Method
mcvol	262.250	ml/mol	McGowan Method
pc	1323.28	kPa	Joback Method
rinpol	2111.00		NIST Webbook
tb	754.32	K	Joback Method
tc	953.96	K	Joback Method

## Sources

Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R159356&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R159356&amp;Units=SI</a>

## Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l

<b>logP:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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