

# Butanoic acid, 3-chloro, 3-methylbutyl ester

<b>Other names:</b>	3-Methylbutyl 3-chlorobutyrate
<b>Inchi:</b>	InChI=1S/C9H17ClO2/c1-7(2)4-5-12-9(11)6-8(3)10/h7-8H,4-6H2,1-3H3
<b>InchiKey:</b>	MRDAYXFMDDPCQA-UHFFFAOYSA-N
<b>Formula:</b>	C9H17ClO2
<b>SMILES:</b>	CC(C)CCOC(=O)CC(C)Cl
<b>Mol. weight [g/mol]:</b>	192.68
<b>CAS:</b>	62108-79-6

## Physical Properties

Property code	Value	Unit	Source
chl	-5340.50	kJ/mol	NIST Webbook
chl	-5350.50 ± 8.40	kJ/mol	NIST Webbook
gf	-225.83	kJ/mol	Joback Method
hf	-589.90 ± 9.60	kJ/mol	NIST Webbook
hfl	-644.30 ± 8.40	kJ/mol	NIST Webbook
hfus	19.00	kJ/mol	Joback Method
hvap	54.40 ± 4.20	kJ/mol	NIST Webbook
log10ws	-2.48		Crippen Method
logp	2.593		Crippen Method
mcvol	157.350	ml/mol	McGowan Method
pc	2363.37	kPa	Joback Method
rinpol	1204.00		NIST Webbook
rinpol	1184.00		NIST Webbook
rinpol	1184.00		NIST Webbook
rinpol	1193.00		NIST Webbook
rinpol	1189.00		NIST Webbook
rinpol	1181.00		NIST Webbook
ripol	1572.00		NIST Webbook
ripol	1586.00		NIST Webbook
ripol	1564.00		NIST Webbook
ripol	1565.00		NIST Webbook
ripol	1572.00		NIST Webbook
tb	518.16	K	Joback Method
tc	704.64	K	Joback Method
tf	263.27	K	Joback Method
vc	0.601	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	352.64	J/molxK	518.16	Joback Method
cpg	366.10	J/molxK	549.24	Joback Method
cpg	378.99	J/molxK	580.32	Joback Method
cpg	391.31	J/molxK	611.40	Joback Method
cpg	403.08	J/molxK	642.48	Joback Method
cpg	414.29	J/molxK	673.56	Joback Method
cpg	424.95	J/molxK	704.64	Joback Method
dvisc	0.0058815	Paxs	263.27	Joback Method
dvisc	0.0023072	Paxs	305.75	Joback Method
dvisc	0.0011372	Paxs	348.23	Joback Method
dvisc	0.0006537	Paxs	390.72	Joback Method
dvisc	0.0004189	Paxs	433.20	Joback Method
dvisc	0.0002907	Paxs	475.68	Joback Method
dvisc	0.0002141	Paxs	518.16	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C62108796&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C62108796&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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