

# 1,2-Distearin

<b>Other names:</b>	(±)-1-(hydroxymethyl)ethane-1,2-diyl distearate
<b>Inchi:</b>	InChI=1S/C39H76O5/c1-3-5-7-9-11-13-15-17-19-21-23-25-27-29-31-33-38(41)43-36-37(
<b>InchiKey:</b>	UHUSDOQQWJGJQS-UHFFFAOYSA-N
<b>Formula:</b>	C39H76O5
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCCCC(=O)OCC(CO)OC(=O)CCCCCCCCCCCCCCCCCCC
<b>Mol. weight [g/mol]:</b>	625.02
<b>CAS:</b>	51063-97-9

## Physical Properties

Property code	Value	Unit	Source
gf	-329.60	kJ/mol	Joback Method
hf	-1495.40	kJ/mol	Joback Method
hfus	102.91	kJ/mol	Joback Method
hvap	137.01	kJ/mol	Joback Method
log10ws	-13.25		Crippen Method
logp	11.957		Crippen Method
mcvol	581.120	ml/mol	McGowan Method
pc	439.50	kPa	Joback Method
tb	1336.04	K	Joback Method
tc	2006.21	K	Joback Method
tf	719.43	K	Joback Method
vc	2.280	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	2257.79	J/molxK	1336.04	Joback Method
cpg	2348.95	J/molxK	1894.52	Joback Method
cpg	2340.44	J/molxK	1782.82	Joback Method
cpg	2329.87	J/molxK	1671.13	Joback Method
cpg	2314.44	J/molxK	1559.43	Joback Method
cpg	2291.34	J/molxK	1447.74	Joback Method
cpg	2358.18	J/molxK	2006.21	Joback Method
dvisc	0.0000002	Paxs	1336.04	Joback Method

dvisc	0.0000003	Paxs	1233.27	Joback Method
dvisc	0.0000005	Paxs	1130.50	Joback Method
dvisc	0.0000008	Paxs	1027.74	Joback Method
dvisc	0.0000017	Paxs	924.97	Joback Method
dvisc	0.0000040	Paxs	822.20	Joback Method
dvisc	0.0000123	Paxs	719.43	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C51063979&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C51063979&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/22-879-2/1-2-Distearin.pdf>

Generated by Cheméo on 2024-04-20 06:41:00.807708608 +0000 UTC m=+15884509.728285919.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.