

# Dichloroacetic acid, 2-pentyl ester

<b>Other names:</b>	Acetic acid, dichloro, 1-methylbutyl ester
<b>Inchi:</b>	InChI=1S/C7H12Cl2O2/c1-3-4-5(2)11-7(10)6(8)9/h5-6H,3-4H2,1-2H3
<b>InchiKey:</b>	BSBLGRPLLFXELO-UHFFFAOYSA-N
<b>Formula:</b>	C7H12Cl2O2
<b>SMILES:</b>	CCCC(C)OC(=O)C(Cl)Cl
<b>Mol. weight [g/mol]:</b>	199.07
<b>CAS:</b>	90380-55-5

## Physical Properties

Property code	Value	Unit	Source
gf	-254.60	kJ/mol	Joback Method
hf	-474.65	kJ/mol	Joback Method
hfus	18.02	kJ/mol	Joback Method
hvap	48.33	kJ/mol	Joback Method
log10ws	-2.64		Crippen Method
logp	2.522		Crippen Method
mcvol	141.410	ml/mol	McGowan Method
pc	2764.26	kPa	Joback Method
rinpol	1102.00		NIST Webbook
ripol	1504.00		NIST Webbook
tb	509.83	K	Joback Method
tc	706.13	K	Joback Method
tf	270.65	K	Joback Method
vc	0.537	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	291.25	J/molxK	509.83	Joback Method
cpg	302.21	J/molxK	542.55	Joback Method
cpg	312.68	J/molxK	575.26	Joback Method
cpg	322.65	J/molxK	607.98	Joback Method
cpg	332.13	J/molxK	640.70	Joback Method
cpg	341.13	J/molxK	673.41	Joback Method

cpg	349.65	J/molxK	706.13	Joback Method
dvisc	0.0052697	Paxs	270.65	Joback Method
dvisc	0.0022910	Paxs	310.51	Joback Method
dvisc	0.0012039	Paxs	350.38	Joback Method
dvisc	0.0007215	Paxs	390.24	Joback Method
dvisc	0.0004754	Paxs	430.10	Joback Method
dvisc	0.0003363	Paxs	469.97	Joback Method
dvisc	0.0002511	Paxs	509.83	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C90380555&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C90380555&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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