

# Pent-4-yn-3-one, 2,2-dimethyl-

<b>Other names:</b>	4,4-Dimethyl-1-pentyne-3-one
<b>Inchi:</b>	InChI=1S/C7H10O/c1-5-6(8)7(2,3)4/h1H,2-4H3
<b>InchiKey:</b>	XRWYQSOYUZOEQH-UHFFFAOYSA-N
<b>Formula:</b>	C7H10O
<b>SMILES:</b>	C#CC(=O)C(C)(C)C
<b>Mol. weight [g/mol]:</b>	110.15
<b>CAS:</b>	5891-25-8

## Physical Properties

Property code	Value	Unit	Source
gf	105.05	kJ/mol	Joback Method
hf	-17.24	kJ/mol	Joback Method
hfus	11.05	kJ/mol	Joback Method
hvap	36.48	kJ/mol	Joback Method
ie	9.75	eV	NIST Webbook
log10ws	-1.59		Crippen Method
logp	1.235		Crippen Method
mcvol	102.460	ml/mol	McGowan Method
pc	3628.97	kPa	Joback Method
tb	400.32	K	Joback Method
tc	602.82	K	Joback Method
tf	267.97	K	Joback Method
vc	0.385	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	191.25	J/molxK	400.32	Joback Method
cpg	202.41	J/molxK	434.07	Joback Method
cpg	212.87	J/molxK	467.82	Joback Method
cpg	222.67	J/molxK	501.57	Joback Method
cpg	231.84	J/molxK	535.32	Joback Method
cpg	240.42	J/molxK	569.07	Joback Method
cpg	248.45	J/molxK	602.82	Joback Method

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5891258&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5891258&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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