

# Glutaric acid, isobutyl 2-isopropoxyphenyl ester

Inchi:	InChI=1S/C18H26O5/c1-13(2)12-21-17(19)10-7-11-18(20)23-16-9-6-5-8-15(16)22-14(3)-
InchiKey:	UFXBHOHVQLFRST-UHFFFAOYSA-N
Formula:	C18H26O5
SMILES:	CC(C)COC(=O)CCCC(=O)Oc1ccccc1OC(C)C
Mol. weight [g/mol]:	322.40

## Physical Properties

Property code	Value	Unit	Source
gf	-374.26	kJ/mol	Joback Method
hf	-822.17	kJ/mol	Joback Method
hfus	35.74	kJ/mol	Joback Method
hvap	78.55	kJ/mol	Joback Method
log10ws	-4.41		Crippen Method
logp	3.749		Crippen Method
mcvol	261.470	ml/mol	McGowan Method
pc	1551.22	kPa	Joback Method
rinpol	2220.00		NIST Webbook
rinpol	2220.00		NIST Webbook
tb	817.02	K	Joback Method
tc	1021.36	K	Joback Method
tf	468.11	K	Joback Method
vc	0.990	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	791.08	J/molxK	817.02	Joback Method
cpg	857.24	J/molxK	987.30	Joback Method
cpg	846.33	J/molxK	953.25	Joback Method
cpg	834.26	J/molxK	919.19	Joback Method
cpg	821.03	J/molxK	885.13	Joback Method
cpg	806.64	J/molxK	851.08	Joback Method
cpg	867.00	J/molxK	1021.36	Joback Method
dvisc	0.0000459	Paxs	817.02	Joback Method

dvisc	0.0000603	Paxs	758.87	Joback Method
dvisc	0.0000830	Paxs	700.72	Joback Method
dvisc	0.0001211	Paxs	642.57	Joback Method
dvisc	0.0001903	Paxs	584.41	Joback Method
dvisc	0.0003307	Paxs	526.26	Joback Method
dvisc	0.0006591	Paxs	468.11	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U358570&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U358570&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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