

# 1,2-Dichlorohexafluoropropane

<b>Other names:</b>	1,2-Dichloro-1,1,2,3,3,3-hexafluoropropane Propane, 1,2-dichloro-1,1,2,3,3,3-hexafluoro- Freon 216 Propane, 1,2-dichlorohexafluoro- Ucon 216 1,1,1,2,3,3-Hexafluoro-2,3-dichloropropane 1,2-Dichloroperfluoropropane
<b>Inchi:</b>	InChI=1S/C3Cl2F6/c4-1(6,2(5,7)8)3(9,10)11
<b>InchiKey:</b>	JSEUKVSKOHVLOV-UHFFFAOYSA-N
<b>Formula:</b>	C3Cl2F6
<b>SMILES:</b>	FC(F)(F)C(F)(Cl)C(F)(F)Cl
<b>Mol. weight [g/mol]:</b>	220.93
<b>CAS:</b>	661-97-2

## Physical Properties

Property code	Value	Unit	Source
chl	-925.60 ± 2.00	kJ/mol	NIST Webbook
gf	-1209.82	kJ/mol	Joback Method
hf	-1349.00 ± 4.40	kJ/mol	NIST Webbook
hfl	-1376.00 ± 4.40	kJ/mol	NIST Webbook
hfus	8.16	kJ/mol	Joback Method
hvap	26.90 ± 0.10	kJ/mol	NIST Webbook
hvap	27.30	kJ/mol	NIST Webbook
log10ws	-3.32		Crippen Method
logp	3.285		Crippen Method
mcvol	88.230	ml/mol	McGowan Method
pc	3110.57	kPa	Joback Method
tb	308.00	K	NIST Webbook
tb	307.30	K	NIST Webbook
tc	446.00	K	NIST Webbook
tf	194.21	K	Joback Method
tt	145.50 ± 0.60	K	NIST Webbook
vc	0.377	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	182.84	J/mol×K	407.51	Joback Method
cpg	189.07	J/mol×K	433.74	Joback Method
cpg	194.73	J/mol×K	459.97	Joback Method
cpg	160.41	J/mol×K	328.83	Joback Method
cpg	168.54	J/mol×K	355.06	Joback Method
cpg	176.01	J/mol×K	381.29	Joback Method
cpg	199.85	J/mol×K	486.20	Joback Method
hvapt	26.28	kJ/mol	307.30	NIST Webbook
hvapt	28.10	kJ/mol	301.50	NIST Webbook
hvapt	25.90 ± 0.10	kJ/mol	313.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C661972&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C661972&amp;Units=SI</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure

<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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