

# Phthalic acid, hexyl 2-tert-butyl-6-methylphenyl ester

<b>Inchi:</b>	InChI=1S/C25H32O4/c1-6-7-8-11-17-28-23(26)19-14-9-10-15-20(19)24(27)29-22-18(2)1
<b>InchiKey:</b>	RPIODNAHQQRKMFV-UHFFFAOYSA-N
<b>Formula:</b>	C25H32O4
<b>SMILES:</b>	CCCCCOC(=O)c1ccccc1C(=O)Oc1c(C)cccc1C(C)(C)C
<b>Mol. weight [g/mol]:</b>	396.52

## Physical Properties

Property code	Value	Unit	Source
gf	-109.45	kJ/mol	Joback Method
hf	-619.03	kJ/mol	Joback Method
hfus	45.58	kJ/mol	Joback Method
hvap	94.80	kJ/mol	Joback Method
log10ws	-7.68		Crippen Method
logp	6.249		Crippen Method
mcvol	330.470	ml/mol	McGowan Method
pc	1197.30	kPa	Joback Method
rinpol	2733.00		NIST Webbook
tb	989.05	K	Joback Method
tc	1218.67	K	Joback Method
tf	608.65	K	Joback Method
vc	1.256	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1075.14	J/molxK	989.05	Joback Method
cpg	1089.64	J/molxK	1027.32	Joback Method
cpg	1102.75	J/molxK	1065.59	Joback Method
cpg	1114.55	J/molxK	1103.86	Joback Method
cpg	1125.10	J/molxK	1142.13	Joback Method
cpg	1134.48	J/molxK	1180.40	Joback Method
cpg	1142.76	J/molxK	1218.67	Joback Method
dvisc	0.0002059	Paxs	608.65	Joback Method
dvisc	0.0001187	Paxs	672.05	Joback Method

dvisc	0.0000752	Paxs	735.45	Joback Method
dvisc	0.0000513	Paxs	798.85	Joback Method
dvisc	0.0000370	Paxs	862.25	Joback Method
dvisc	0.0000279	Paxs	925.65	Joback Method
dvisc	0.0000218	Paxs	989.05	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357097&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357097&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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