

# 2,5-Dichlorobenzyl alcohol, methyl ether

<b>Inchi:</b>	InChI=1S/C8H8Cl2O/c1-11-5-6-4-7(9)2-3-8(6)10/h2-4H,5H2,1H3
<b>InchiKey:</b>	KULCZQDEZITVPV-UHFFFAOYSA-N
<b>Formula:</b>	C8H8Cl2O
<b>SMILES:</b>	COCc1cc(Cl)ccc1Cl
<b>Mol. weight [g/mol]:</b>	191.06

## Physical Properties

Property code	Value	Unit	Source
gf	-19.23	kJ/mol	Joback Method
hf	-158.56	kJ/mol	Joback Method
hfus	19.32	kJ/mol	Joback Method
hvap	48.18	kJ/mol	Joback Method
log10ws	-3.23		Crippen Method
logp	3.140		Crippen Method
mcvol	130.170	ml/mol	McGowan Method
pc	3170.40	kPa	Joback Method
rinpol	1343.00		NIST Webbook
rinpol	1343.00		NIST Webbook
tb	516.36	K	Joback Method
tc	739.42	K	Joback Method
tf	313.45	K	Joback Method
vc	0.491	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	247.81	J/molxK	516.36	Joback Method
cpg	258.26	J/molxK	553.54	Joback Method
cpg	268.15	J/molxK	590.71	Joback Method
cpg	277.50	J/molxK	627.89	Joback Method
cpg	286.30	J/molxK	665.07	Joback Method
cpg	294.58	J/molxK	702.24	Joback Method
cpg	302.33	J/molxK	739.42	Joback Method
dvisc	0.0013663	Paxs	313.45	Joback Method

dvisc	0.0008682	Paxs	347.27	Joback Method
dvisc	0.0005980	Paxs	381.09	Joback Method
dvisc	0.0004376	Paxs	414.90	Joback Method
dvisc	0.0003357	Paxs	448.72	Joback Method
dvisc	0.0002673	Paxs	482.54	Joback Method
dvisc	0.0002193	Paxs	516.36	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U378114&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U378114&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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