

# cis-1,3-Diethenylcyclobutane

<b>Inchi:</b>	InChI=1S/C8H12/c1-3-7-5-8(4-2)6-7/h3-4,7-8H,1-2,5-6H2/t7-,8+
<b>InchiKey:</b>	AATSALLMQGUKIE-OCAPTIKFSA-N
<b>Formula:</b>	C8H12
<b>SMILES:</b>	C=CC1CC(C=C)C1
<b>Mol. weight [g/mol]:</b>	108.18
<b>CAS:</b>	77614-53-0

## Physical Properties

Property code	Value	Unit	Source
gf	233.10	kJ/mol	Joback Method
hf	88.71	kJ/mol	Joback Method
hfus	11.02	kJ/mol	Joback Method
hvap	31.84	kJ/mol	Joback Method
ie	9.38	eV	NIST Webbook
log10ws	-2.29		Crippen Method
logp	2.385		Crippen Method
mcvol	104.120	ml/mol	McGowan Method
pc	3156.17	kPa	Joback Method
tb	382.14	K	Joback Method
tc	574.70	K	Joback Method
tf	186.58	K	Joback Method
vc	0.394	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	185.55	J/molxK	382.14	Joback Method
cpg	200.19	J/molxK	414.23	Joback Method
cpg	214.06	J/molxK	446.33	Joback Method
cpg	227.19	J/molxK	478.42	Joback Method
cpg	239.60	J/molxK	510.51	Joback Method
cpg	251.33	J/molxK	542.61	Joback Method
cpg	262.41	J/molxK	574.70	Joback Method
dvisc	0.0008087	Paxs	186.58	Joback Method

dvisc	0.0005958	Paxs	219.17	Joback Method
dvisc	0.0004751	Paxs	251.77	Joback Method
dvisc	0.0003990	Paxs	284.36	Joback Method
dvisc	0.0003474	Paxs	316.95	Joback Method
dvisc	0.0003104	Paxs	349.55	Joback Method
dvisc	0.0002827	Paxs	382.14	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C77614530&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C77614530&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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