

# Benzoyl chloride, 3-chloro-

<b>Other names:</b>	3-Chlorobenzoyl chloride Benzoyl chloride, m-chloro- m-Chlorobenzoyl chloride
<b>Inchi:</b>	InChI=1S/C7H4Cl2O/c8-6-3-1-2-5(4-6)7(9)10/h1-4H
<b>InchiKey:</b>	WHIHIKVIWVIER-UHFFFAOYSA-N
<b>Formula:</b>	C7H4Cl2O
<b>SMILES:</b>	O=C(Cl)c1cccc(Cl)c1
<b>Mol. weight [g/mol]:</b>	175.01
<b>CAS:</b>	618-46-2

## Physical Properties

Property code	Value	Unit	Source
gf	-41.94	kJ/mol	Joback Method
hf	-106.81	kJ/mol	Joback Method
hfl	-190.00 ± 1.60	kJ/mol	NIST Webbook
hfus	17.53	kJ/mol	Joback Method
hvap	49.63	kJ/mol	Joback Method
log10ws	-3.05		Crippen Method
logp	2.719		Crippen Method
mcvol	111.780	ml/mol	McGowan Method
pc	3980.54	kPa	Joback Method
rinpol	1246.60		NIST Webbook
rinpol	1246.60		NIST Webbook
tb	498.20	K	NIST Webbook
tc	758.29	K	Joback Method
tf	317.36	K	Joback Method
vc	0.423	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	239.57	J/mol×K	758.29	Joback Method
cpg	233.90	J/mol×K	718.56	Joback Method
cpg	227.70	J/mol×K	678.84	Joback Method

cpg	220.95	J/mol×K	639.12	Joback Method
cpg	213.61	J/mol×K	599.40	Joback Method
cpg	205.65	J/mol×K	559.67	Joback Method
cpg	197.03	J/mol×K	519.95	Joback Method
dvisc	0.0021445	Paxs	317.36	Joback Method
dvisc	0.0003235	Paxs	519.95	Joback Method
dvisc	0.0003974	Paxs	486.19	Joback Method
dvisc	0.0005035	Paxs	452.42	Joback Method
dvisc	0.0006626	Paxs	418.66	Joback Method
dvisc	0.0009150	Paxs	384.89	Joback Method
dvisc	0.0013446	Paxs	351.12	Joback Method
hvapt	49.40	kJ/mol	379.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.36924e+01
Coeff. B	-4.29827e+03
Coeff. C	-4.80940e+01
Temperature range (K), min.	368.75
Temperature range (K), max.	560.96

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C618462&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C618462&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

**cpg:** Ideal gas heat capacity

<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/23-763-9/Benzoyl-chloride-3-chloro.pdf>

Generated by Cheméo on 2024-04-23 07:09:14.849493047 +0000 UTC m=+16145403.770070358.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.